(11) EP 1 234 569 A1

(12)

EUROPEAN PATENT APPLICATION published in accordance with Art. 158(3) EPC

- (43) Date of publication: 28.08.2002 Bulletin 2002/35
- (21) Application number: 00978057.8
- (22) Date of filing: 01.12.2000

- (51) Int Cl.7: **A61 K 7/13**, C09B 53/00, C09B 55/00, C09B 57/00
- (86) International application number: PCT/€P00/08525
- (87) International publication number: WO 01/039736 (07.06.2001 Gazette 2001/23)
- (84) Designated Contracting States:

 AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU

 MC NL PT SE TR

 Designated Extension States:

 AL LT LV MK RO SI
- (30) Priority: 02.12.1999 JP 34393099
- (71) Applicant: Lion Corporation Tokyo 130-8644 (JP)
- (72) Inventors:
 - MITAMURA, Joji
 Sumida-ku, Tokyo 130-8644 (JP)

- NOGUCHI, Mutsumi Sumida-ku, Tokyo 130-8644 (JP)
- ONUKI, Takeshi Sumida-ku, Tokyo 130-8644 (JP)
- (74) Representative: Stuart, Ian Alexander et al MEWBURN ELLIS York House
 23 Kingsway
 London WC2B 6HP (GB)

(54) COMPOSITIONS FOR DYEING HORNY FIBERS

(57) A composition for dyeing keratinous fiber which includes incorporated therein an oxidative color-developing substance, an enzyme which reacts with oxygen as a substrate but does not evolve hydrogen peroxide, and a weak reducing agent.

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Description

TECHNICAL FIELD

[0001] The present invention relates to a composition for dyeing keratinous fiber which comprises an oxidative color-developing substance, an enzyme which reacts with oxygen as a substrate but does not evolve hydrogen peroxide, and a weak reducing agent. More particularly, the present invention relates to a composition for dyeing keratinous fiber which is well protected from discoloration of compositions with time.

10 BACKGROUND ART

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[0002] There are compositions for dyeing keratinous fiber which perform their function by oxidation of an oxidative color-developing substance, which include, for example, hair dyes and eyebrow dyes. Such compositions for dyeing keratinous fiber conventionally employ hydrogen peroxide as an oxidizing agent. Therefore, most of them are of two-pack type. That is, the oxidative color-developing agent and the oxidizing agent are stored in separate containers, and they are mixed together for reaction at the time of use. The composition of this type is inconvenient to use, and there has been a demand for improvement in their usability. Moreover, it is known that hydrogen peroxide damages keratinous fiber such as hair, and this has produced dissatisfaction with consumers.

[0003] One way to address the above-mentioned problem is to replace hydrogen peroxide by an oxidase (as an oxidizing agent) which is previously mixed with an oxidative color-developing substance. The oxidase may be peroxidase (as disclosed in Japanese Patent Laid-open Nos. Sho 47-10400 and Sho 53-32132), laccase (as disclosed in U.S. Patent No. 3251742 and Japanese Patent Laid-open No. Hei 6-172145), or uricase (as disclosed in Japanese Patent Laid-open No. Sho 63-246313).

[0004] Among these disclosed technologies, however, the peroxidase-containing composition needs hydrogen peroxide on account of the characteristic properties of peroxidase, and hence the resulting composition cannot be of one-pack type. The uricase-containing composition can be of one-pack type, but it does not solve the basic problem with the use of hydrogen peroxide evolved by the enzyme reaction.

[0005] By contrast, in the case where an oxidase which reacts with oxygen as a substrate but does not evolve hydrogen peroxide is used, the composition for dyeing keratinous fiber can be of one-pack type. In addition, such a composition is useful because it does not evolve hydrogen peroxide which damages keratinous fiber (See Japanese Patent Laid-open No. Hei 11-60454). However, it suffers the disadvantage of getting discolored with time during storage, because the enzyme (as a protein) is unstable under the high-temperature high-humidity condition. Therefore, the resulting composition does not permit the enzyme to fully perform its function, and its discoloration is a serious drawback.

[0006] It has been known that the above-mentioned discoloration can be effectively eliminated by incorporation with a reducing agent (See "Science of Wave" issued by Shinbiyo Shuppan Co., Ltd.). Improvement of storage stability of enzyme is disclosed in Japanese Patent Laid-open No. Hei 8-175935 (for catalase) and Japanese Patent Laid-open No. Hei 8-217652 (for uricase). These disclosed technologies employ a reducing agent, however, incorporation of a reducing agent into the composition for dyeing keratinous fiber poses a problem. That is, a strong reducing agent lowers the enzyme activity, thereby deteriorating dyeing power, and a weak reducing agent does not fully protect the composition from discoloration.

DISCLOSURE OF INVENTION

[0007] The present invention was completed in view of the foregoing, and it is accordingly an object of the present invention to provide a composition for dyeing keratinous fiber which does not damage keratinous fiber, protects itself from discoloration during storage, exhibits good enzyme action, and allows consumers easy use.

[0008] The present inventors' investigation to achieve the above-mentioned object revealed that if a composition for dyeing keratinous fiber, which is composed of an oxidative color-developing substance and an enzyme which reacts with oxygen as a substrate but does not evolve hydrogen peroxide, is incorporated with a weak reducing agent, the resulting composition has greatly improved stability with time, as demonstrated in the following Examples. That is, the composition protects itself from discoloration with time (particularly during storage at high temperatures) without deteriorating dyeing power, and remains free from insoluble aggregates during storage. The investigation revealed further that the composition has better storage stability upon incorporation with cyclodextrin. The present invention is based on these findings.

[0009] Accordingly, the present invention is directed to a composition for dyeing keratinous fiber which comprises incorporated therein an oxidative color-developing substance, an enzyme which reacts with oxygen as a substrate but does not evolve hydrogen peroxide, and a weak reducing agent. The composition should preferably be incorporated

with cyclodextrin.

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BEST MODE FOR CARRYING OUT THE INVENTION

[0010] In what follows, the invention will be described in more detail. According to the present invention, the composition for dyeing keratinous fiber comprises incorporated therein an oxidative color-developing substance, an enzyme which reacts with oxygen as a substrate but does not evolve hydrogen peroxide, and a weak reducing agent. It should preferably be additionally incorporated with cyclodextrin.

[0011] According to the present invention, the composition for dyeing keratinous fiber is not restricted in the form as a commodity, but it may also be in the form of reactive hair dye or reactive dye for eyebrows, eyelashes, and body hair. Regardless of the commodity type, the composition should be a solution or an emulsion (which may contain a propellant) in which the enzyme is dissolved. The composition at the time of use may be in the form of foam, cream, or clear gel. [0012] According to the present invention, the oxidative color-developing substance is not specifically restricted in its kind, but Colorant precursors, developers, and direct dyes may also be used. It includes all ordinary oxidation dyes, such as those listed in the standard for raw materials of hair dyes (The 3rd revised edition, issued in May 1985 by Japan Hair Color Industry Association, Japan Hair Dye Industry Conference). Their typical examples are as follows. 5-amino-o-cresol, o-aminophenol, m-aminophenol, p-aminophenol, 2,6-diaminopyridine, 5-(2-hydroxyethylamino)-2-methylphenol, N,N-bis(2-hydroxyethyl)-p-phenylenediamine, p-nitro-o-phenylenediamine, p-phenylenediamine, mphenylenediamine, N-phenyl-p-phenylenediamine, catechol, resorcin, hydroquinone, 3,3'-iminodiphenol, diphenylamine, 2-hydroxy-5-nitro-2',4'-diaminobenzene'sodium sulfate, toluene-2,5-diamine, 2-(2'-hydroxyethylamino)-5-aminotoluene, N,N-bis(2-hydroxyethyl)-p-phenylenediamine sulfate, 5-amino-o-cresol sulfate, p-aminophenol sulfate, o-chloro-p-phenylenediamine sulfate, 2-(2'-hydroxyethylamine)-5-aminotoluene sulfate, 4,4'-diaminodiphenylamine sulfate, p-methylaminophenyl sulfate, p-phenylenediamine sulfate, m-phenylenediamine sulfate, toluene-2,5-diamine sulfate, 2,4-diaminophenoxyethanol hydrochloride, toluene-2,5-diamine hydrochloride, m-phenylenediamine hydrochloride, 2,4-diaminophenyl hydrochloride, N-phenyl-p-phenylenediamine hydrochloride, 2,4-diaminophenol hydrochloride, N-phenyl-p-phenylenediamine hydrochloride, N-phenyl-p-phenylenediamine acetate, 1,5-hyroxynaphthalene, toluene-3,4-diamine, p-methylaminophenol, N,N'-bis(4-aminophenyl)-2,5-diamino-1,4-quinonedimine, o-aminophenol sulfate, 2,4-diaminophenol sulfate, m-aminophenol sulfate, 2-amino-4-nitrophenol, 2-amino-5-nitrophenol, 1-amino-4-methylaminoanthraquinone, nitro-p-phenylenediamine hydrochloride, 1,4-diaminoanthraquinone, nitro-p-phenylenediamine, (-naphthol, 1,5-dihydroxynaphthelene, pyrogallol, phloroglucin, picric acid, picramic acid, sodium picramate, p-aminophenylsulfamic acid, 2-amino-5-nitrophenol sulfate, nitro-p-phenylenediamine sulfate, p-nitro-o-phenylenediamine sulfate, and p-nitro-m-phenylenediamine sulfate.

Of these examples, the following are particularly preferable:

p-phenylenediamine or salt thereof, toluene-2,5-diamine or salt thereof, p-aminophenol, 5-amino-o-cresol, p-methylaminophenol, sulfate, m-aminophenol, p-nitro-o-phenylenediamine, 2,6-diaminopyridine, resorcinol, o-aminophenol, and m-phenylenediamine.

[0013] In addition, it is also possible to use various oxidation dyes (new compounds or oxidative base compounds) and couplers which are disclosed in the following international applications filed by Loreal Co., Ltd. They are listed below for reference:

WO99/36034, WO99/36035, WO99/36036, WO99/36037, WO99/36038, WO99/36039, WO99/36040, WO99/36041 WO99/36042, WO99/36043, WO99/36044, WO99/36045, and WO99/36046.

[0014] Examples of the oxidative base compounds include paraphenylenediamine, double base compounds, p-aminophenols, o-aminophenols, and heterocyclic oxidative base compounds.

[0015] The p-phenylenediamine which is desirable as the oxidative base compound to be incorporated into the composition of the present invention includes those compounds and acid adducts (or acid salts) thereof which are represented by the following general formula (1).

$$R^4$$
 NR^1R^2
 R^3
 NH_2

[0016] R¹ denotes a hydrogen atom, C1-4 alkyl group, C1-4 monohydroxyalkyl group, C2-4 polyhydroxyalkyl group, (C1-4)alkoxy(C1-4)alkyl group, and C1-4 alkyl group substituted with a nitrogen-containing group, phenyl group, or 4'-aminophenyl group; R² denotes a hydrogen atom, C1-4 alkyl group, C1-4 monohydroxyalkyl group, C2-4 polyhydroxyalkyl group. (C1-4)alkoxy(C1-4)alkyl group, and C1-4 alkyl group substituted with a nitrogen-containing group; R³ denotes a hydrogen atom, halogen atom (such as chlorine, bromine, iodine, and fluorine), C1-4 alkyl group, C1-4 hydroxyalkyl group, C1-4 hydroxyalkoxyl group, C1-4 acetylaminoalkoxyl group, C1-4 mesylaminoalkoxyl group, and C1-4 carbamoylaminoalkoxyl group; and R⁴ denotes a hydrogen atom, halogen atom, or C1-4 alkyl group.

[0017] The nitrogen-containing group in the general formula (1) includes, for example, amino group, mono(C1-4) alkylamino group, di(C1-4)alkylamino group, tri(C1-4)alkylamino group, monohydroxy(C1-4)alkylamino group, imidazolinium, and ammonium.

[0018] Typical examples of the paraphenylenediamine in the general formula (1) above include the following: Paraphenylenediamine, paratolenediamine, 2-chloroparaphenyenediamine, 2,3-dimethylparaphenylenediamine, 2,6-dimethylparaphenylenediamine, 2,5-dimethylparaphenylenediamine, N,N-dimethylparaphenylenediamine, N,N-dipropylparaphenylenediamine, N,N-dipropylparaphenylenediamine, 4-amino-N,N-diethyl-3-methylaniline, N,N-bis(β -hydroxyethyl)paraphenylenediamine, 4-N,N-bis(β -hydroxyethyl)amino-2-methylaniline, 4-N,N-bis(β -hydroxyethyl)amino-2-chloroaniline, 2- β -hydroxyethylparaphenylenediamine, 2-fluoroparaphenylenediamine, 2-isopropylparaphenylenediamine, N-(β -hydroxypropyl)paraphenylenediamine, 2-hydroxymethylparaphenylenediamine, N,N-dimethyl-4-methylparaphenylenediamine, N,N-(ethyl- β -hydroxyethyl)paraphenylenediamine, N-(β -methoxyethyl)paraphenylenediamine, N-(β -acetylaminoethyloxyparaphenylenediamine, N-(β -methoxyethyl)paraphenylenediamine, and salts thereof.

[0019] Of the paraphenylenediamines in the general formula (1) above, those listed below are particularly preferable: Paraphenylenediamine, paratolylenediamine, 2-isopropylparaphenylenediamine, 2- β -hydroxyethylparaphenylenediamine, 2- β -hydroxyethyloxyparaphenylenediamine, 2,6-dimethylparaphenylenediamine, 2,3-dimethylparaphenylenediamine, N,N-bis(β -hydroxyethyl)paraphenylenediamine, 2-chloroparaphenylenediamine, 2- β -acetylaminoethylparaphenylenediamine, and salts thereof.

[0020] The term "double base compound" as used in the present invention implies any compound which contains at least two aromatic rings which are cross-linked by an amino group and/or hydroxyl group.

[0021] The double base compound that can be incorporated into the composition of the present invention includes the compounds and acid salts thereof represented by the general formula (2) below.

[0022] Z^1 and Z^2 (which may be identical or different) each denotes a hydrogen atom, hydroxyl group, or amino group, which may be substituted with a C1-4 alkyl group or a cross-linking group Y. The cross-linking group Y is a C1-14 linear or branched alkylene group, which may have one or more nitrogen-containing groups and/or one or more heteroatoms (such as oxygen, sulfur, and nitrogen) interposed or substituted. In addition, it may be substituted with one or more hydroxyl groups or C1-6 alkoxyl groups.

[0023] R⁵ and R⁶ each denotes any of hydrogen atom, halogen atom, C1-4 alkyl group, C1-4 monohydroxyalkyl group, C2-4 polyhydroxyalkyl group, C1-4 aminoalkyl group, or cross-linking group Y; R⁷, R⁸, R⁹, R¹⁰, R¹¹ (which may be identical or different) each denotes a hydrogen atom, cross-linking group Y, or C1-4 alkyl group.)

[0024] It is assumed that the compound represented by the general formula (2) above has one cross-linking group per molecule.

[0025] The nitrogen-containing group in the general formula (2) above includes, for example, an amino group, mono (C1-4)alkylamino group, di(C1-4)alkylamino group, tri(C1-4)alkylamino group, monohydroxy(C1-4)alkylamino group, imidazolinium, and ammonium.

[0026] The double base compound in the general formula (2) above is exemplified by the following: $N,N=bis-(\beta-hydroxyethyl)N,N'=bis-(4'-aminophenyl)-1,3$ diaminopropanol, $N,N=bis-(\beta-hydroxyethyl)N,N'=bis-(4'-aminophenyl)$ ethylenediamine, $N,N=bis-(\beta-hydroxyethyl)-N,N'-bis-(4-aminophenyl)$ tetramethylenediamine, $N,N'-bis-(\beta-hydroxyethyl)-N,N'-bis-(4-aminophenyl)$

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nyl)tetramethylenediamine, N,N'-bis-(4-methylaminophenyl)tetramethylenediamine, N,N'-bis-(ethyl)N,N'-bis-(4-amino 3-methylphenyl)ethylenediamine, 1,8-bis-(2,5-diaminophenoxy)-3,5-dioxaoctane, and acid salts thereof.

[0027] Of the above-mentioned examples in the general formula (2), the following double base compounds are particularly preferable:

N,N-bis-(β-hydroxyethyl)N,N'-bis-(4'-aminophenyl)1,3 diaminopropanol, 1,8-bis-(2,5-diaminophenoxy)-3,5-dioxaoctane, and acid salts thereof.

[0028] The oxidative base compound (or p-aminophenol) that can be incorporated into the composition of the present invention includes those compounds and acid salts thereof which are represented by the general formula (3) below.

$$\begin{array}{c}
\text{OH} \\
\mathbb{R}^{13}
\end{array}$$

$$\begin{array}{c}
\mathbb{R}^{14}
\end{array}$$

$$\begin{array}{c}
\mathbb{R}^{14}
\end{array}$$

[0029] R¹³ denotes a hydrogen atom, halogen atom, C1-4 alkyl group, C1-4 monohydroxyalkyl group, (C1-4)alkoxyl group, C1-4 aminoalkyl group, or (C1-4)hydroxyalkyl(C1-4)aminoalkyl group; R¹⁴ denotes a hydrogen atom, halogen atom, C1-4 alkyl group, C1-4 monohydroxyalkyl group, C2-4 polyhydroxyalkyl group, C1-4 aminoalkyl group, C1-4 cyanoalkyl group, or (C1-4)alkoxy(C1-4)alkyl group. At least either of R¹³ and R¹⁴ is a hydrogen atom.

[0030] Typical examples of the p-aminophenol in the general formula (3) above includes the following: p-aminophenol, 4-amino 3-methylphenol, 4-amino 3-fluorophenol, 4-amino 3-hydroxymethylphenol, 4-amino 2-methylphenol, 4-amino 2-hydroxymethylphenol, 4-amino 2-methoxyphenol, 4-amino 2-aminophenol, 4-amino 2-(β-hydroxyethylaminoethyl)phenol, 4-mino 2-fluorophenol, and acid salts thereof.

[0031] The oxidative base compound (or o-aminophenol) that can be incorporated into the composition of the present invention includes, for example, 2-aminophenol, 2-amino 5-methylphenol, 2-amino 6-methylphenol, 5-acetamide 2-aminophenol, and acid salts thereof.

[0032] The oxidative base compound (or heterocyclic base compound) that can be incorporated into the composition of the present invention includes, for example, pyridine derivative, pyrimidine derivatives, pyrazole derivatives, pyrazole pyrimidine derivatives, and acid salts thereof. The above-mentioned pyridine derivatives includes those compounds mentioned in GB-PS1026978 and GB-PS1153196, such as 2,5-diaminopyridine, 2-(4-methoxyphenyl)amino 3-aminopyridine, 2,3-diamino 6-methoxypyridine, 2-(β-methoxyethyl)amino 3-amino 6-methoxypyridine, 3,4-diaminopyridine, and acid salts thereof.

[0033] Examples of the above-mentioned pyrimidine derivatives include those compounds mentioned in German Patent No. DE2359399, Japanese Patent Nos. JP88-169571 and JP91-333495, and International Laid-open No. WO96/15765, such as 2,4,5,6-tetraaminopyridine, 4-hydroxy 2,5,6-triaminopyridine, 2-hydroxy 4,5,6-triaminopyridine, 2,4-dihydroxy 5,6-diaminopyridine, 2,5,6-triaminopyridine, and acid salts thereof.

[0034] Examples of the above-mentioned pyrazole derivatives include those compounds mentioned in German Patent Nos. DE3843892, DE4133957, DE19543988, International Laid-open Nos. WO94/08969 and WO94/08970, and French Patent No. FR-A-2733, as they are listed below:

4,5-diamino 1-methylpyrazole, 3,4-diaminopyrazole, 4,5-diamino 1-(4'-chlorobenzylpyrazole), 4,5-diamino 1,3-dimethylpyrazole, 4,5-diamino 3-methyl-1-phenylpyrazole, 4,5-diamino 1-methyl 3-phenylpyrazole, 4-amino 1,3 dimethyl 5-hydrazionopyrazole, 1-benzyl 4,5-diamino 3-methylpyrazole, 4,5-diamono 3-tert-butyl-1- methylpyrazole, 4,5-diamono 1-tert-butyl 3-methylpyrazole, 4,5-diamino 1-(β-hydroxyethyl)-3-methylpyrazole, 4,5-diamino 1-ethyl 3-methylpyrazole, 4,5-diamino 3-hydroxymethyl 1-methylpyrazole, 4,5-diamino 3-hydroxymethyl 1-methylpyrazole, 4,5-diamino 3-hydroxymethyl 1-isopropylpyrazole, 4,5-diamino 3-methyl 1-isopropylpyrazole, 4-amino 5-(2'-aminoethyl)amino 1,3-dimethylpyrazole, 3,5-diamino 1-methyl 4-methylaminopyrazole, 3,5-diamino 4-(β-hydroxyethyl) amino 1-methylpyrazole, and acid salts thereof.

[0035] Examples of the above-mentioned pyrazolopyrimidine derivatives include pyrazole-[1,5,a]-pyrimidine derivative or acid salt thereof which is represented by the general formula (4) below. The derivatives include tautomers thereof if there exist tautomenc equilibrium.

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$$(X)_{i} = \underbrace{ \begin{cases} N \\ N \end{cases} }_{5} \underbrace{ \begin{cases} N \\ N \end{cases} }_{7} \underbrace{ \begin{cases} N \\ N \end{cases} }_{2} \underbrace{ \begin{cases} N \\ N \end{cases} }_{7} \underbrace$$

[0036] R15, R16, R17, and R18 (which may be identical or different) each denotes a hydrogen atom, C1-4 alkyl group, aryl group, C1-4 hydroxyalkyl group, C2-4 polyhydroxyalkyl group; (C7-4)alkoxy(C1-4)alkyl group, C1-4 aminoalkyl group (in which the amino group may be protected with an acetyl group, ureido group, or sulfonyl group), (C1-4)alkylamino(C1-4)alkyl group, di[(C1-4)alkyl]amino(C1-4)alkyl group (in which the dialkyl group may form a cyclic hydrocarbon group, 5- or 6-membered heterocyclic group), hydroxy(C1-4)alkyl group, or di[hydroxy(C1-4)alkyl]amino(C1-4)alkyl group.

[0037] The X groups (which may be identical or different) each denotes a hydrogen atom, C1-4 alkyl group, aryl group, C1-4 hydroxyalkyl group, C2-4 polyhydroxyalkyl group, amino(C1-4)alkyl group, (C1-4)alkyl group, (C1-4)alkyl group, di[(C1-4)alkyl]amino(C1-4)alkyl group (in which the dialkyl group may form a cyclic hydrocarbon group, 5- or 6-membered heterocyclic group), hydroxy(C1-4)alkyl group, di[hydroxy(C1-4)alkyl]amino(C1-4)alkyl group, amino group, di[(C1-4)alkyl]amino(C1-4)alkyl group, halogen atom, carboxylic group, or sulfonic group.

[0038] i denotes 0, 1, or 3; p denotes 0 or 1; q denotes 0 or 1; and n denotes 0 or 1, provided that p+q is not 0. If p+q is 2, n denotes 0 and NR¹5R¹6 and NR¹7R¹8 are at any position of (2,3), (5,6), (6,7), (3,5), and (3,7). If p+q is 1, n denotes 1 and NR¹5R¹6 (or NR¹7R¹8) and the hydroxyl group are at any position of (2,3), (5,6), (6,7), (3,5), and (3,7). [0039] In the case where the pyrazolo-[1,5,a]-pyrimidine derivative represented by the general formula (4) above has a hydroxyl group at any one position of 2, 5, and 7 (which is α position with respect to the nitrogen atom), there exists tautomeric equilibrium as represented by the following reaction formula.

[0040] Examples of the pyrazolo-[1,5,a]-pyrimidine derivative represented by the general formula (4) above include the following:

Pyrazolo-[1,5,a]-pyrimidine-3,7-diamine, 2,5-dimethylpyrazolo-[1,5,a]-pyrimidine-3,7-diamine, pyrazolo-[1,5,a]-pyrimidine-3,5-diamine, 2,7-dimethylpyrazolo-[1,5,a]-pyrimidine-3,5-diamine, 3-aminopyrazolo-[1,5,a]-pyrimidine-7-ol, 3-aminopyrazolo-[1,5,a]-pyrimidine-7-ol, 2-(3-aminopyrazolo-[1,5,a]-pyrimidine-7-ylamino)-ethanol, 2-(7-aminopyrazolo-[1,5,a]-pyrimidine-7-yl)-(2-hydroxy-ethyl)-amino]ethanol, 2-[(7-aminopyrazolo-[1,5,a]-pyrimidine-3-yl)-(2-hydroxyethyl)-amino]ethanol, 5,6-dimethylpyrazolo-[1,5,a]-pyrimidine-3,7-diamine, 2,6-dimethylpyrazolo-[1,5,a]-pyrimidine-3,7-diamine, 2,5-N',N'-teteramethylpyrazolo-[1,5,a]-pyrimidine-3,7-diamine, and acid salts thereof, and tautomers thereof if there exist their tautomeric equilibrium. [0041] The pyrazolo-[1,5,a]-pyrimidine derivative represented by the general formula (4) above can be synthesized by cyclization of aminopyrazole according to the method mentioned in the following literature:

- (1) EP No. 628559, BEIERSDORF-LILLY
- (2) R. Vishdu, H. Navedul, Indian J. Chem., 34b(6), 514, 1995.
- (3) N. S. Ibrahim, K. U. Sadek, F. A. Abdel-Al, Arch. Pharm., 320, 240, 1987.
- (4) R. H. Springeer, M. B. Scholten, D. E. O'Brien, T. Novinson, J. P. Miller, R. K. Robins, J. Med. Chem., 25, 235, 1982.

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- (5) T. Novinson, R. K. Robins, T. R. Matthews, J. Med. Chem., 20, 296, 1977.
- (6) US No. 3907799, ICN PHARMACEUTICAL

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[0042] The pyrazolo-[1,5,a]-pyrimidine derivative represented by the general formula (4) above can also be synthesized by cyclization of hydrazine according to the method mentioned in the following literature:

- (1) A. McKillop, R. J. Kobilecki, Heterocycles, 6(9), 1355, 1977.
- (2) E. Alcade, J. DeMendora, J. M. Macrcia-Marquina, C. Almera, J. Elguero, J. Heterocyclic Chem., 11(3), 423, 1974.
- (3) K. Saito, I. Hori, M. Igarashi, H. Midorikawa, Bull. Chem. Soc. Japan, 47(2), 476, 1974.

[0043] The above-mentioned oxidative base compound should preferably be added in an amount of 0.0005 to 12%, particularly 0.005 to 6%, of the total amount of the composition of the present invention. (% means mass% hereinafter.)
[0044] The coupler that can be added to the composition of the present invention includes m-phenylenediamine, m-aminophenol, m-diphenol, heterocyclic coupler, and acid salts thereof, which are commonly used for oxidation hair dyes.

[0045] Examples of the coupler are listed below. 2-methyl-5-aminophenol, 5-N-(β-hydroxyethyl)-amino-2-methylphenol, 3-aminophenol, 1,3-dihydroxybenzene, 1,3-dihydroxy-2-methylbenzene, 4-chloro-1,3-dihydroxybenzene, 2,4-diamino-1-(β-hydroxyethyloxy)-benzene, 2-amino-4-(β-hydroxyethylamino)-1-methoxybenzene, 1,3-diaminobenzene, 1,3-bis-(2,4-diaminophenoxy)-propane, sesamol, α-naphthol, 6-hydroxyindole, 4-hydroxyindole, 4-hydroxy-N-methylindole, 6-hydroxyindoline, 2,6-dihydroxy-4-methylpyridine, 1-H-3-methylpyrazol-5-on, 1-phenyl-3-methyl-pyrazol-5-on, 2,6-dimethylpyrazole-[1,5b]-1,2,4-triazole, 2,6-dimethyl-[3,2-c]-1,2,4-triazole, 6-methylpyrazolo-[1,5-a]-benzimidazole, 6-methylpyrazolo-[1,5-a]-benzimidazole, and acid salts thereof.

[0046] Examples of the m-aminophenol include those compounds and acid salts thereof represented by the general formula (5) below.

$$R^{21}$$

$$R^{20}$$
NHR¹⁹
(5)

[0047] R^{19} denotes a hydrogen atom, C1-4 alkyl group, C1-4 monohydroxyalkyl group, or C2-4 polyhydroxyalkyl group; R^{20} denotes a hydrogen atom, C1-4 alkyl group, C1-4 alkoxyl group, or halogen atom (which is any of chlorine, iodine, bromine, and fluorine); and R^{21} denotes a hydrogen atom, C1-4 alkyl group, C1-4 alkoxyl group, C1-4 monohydroxyalkyl group, C2-4 polyhydroxyalkyl group, C1-4 monohydroxyalkoxyl group, or C2-4 polyhydroxyalkoxyl group. [0048] Typical examples of the m-aminophenol represented by the general formula (5) above are listed below. m-aminophenol, 5-amino-2-methoxyhenol, 5-amino-2-(β -hydroxyethoxy)-phenol, 5-amino-2-methylphenol, 5-N-(β -hydroxyethyl)amino-2-methylphenol, 5-N-(β -hydroxyethyl)amino-4-methoxy-2-methylphenol, 5-amino-4-chloro-2-methylphenol, 5-amino-2,4-dimethoxyphenol, 5-(γ -hydroxypropylamino)-2-methylphenol, and acid salts thereof.

The m-phenylenediamine used as the coupler in the composition of the present invention should preferably be any compound or acid salt thereof represented by the general formula (6) below.

$$R^{25}$$

$$R^{24}$$

$$R^{23}$$

$$R^{23}$$

$$R^{23}$$

$$R^{23}$$

$$R^{23}$$

$$R^{24}$$

[0049] R²² denotes a hydrogen atom, C1-4 alkyl group, C1-4 monohydroxyalkyl group, or C2-4 polyhydroxyalkyl group; R²³ and R²⁴ (which may be identical or different) each denotes a hydrogen atom, C1-4 alkyl group, C1-4 monohydroxyalkyl group, or C2-4 polyhydroxyalkyl group; and R²⁵ denotes a hydrogen atom, C1-4 alkoxyl group, C1-4 aminoalkoxyl group, C1-4 monohydroxyalkoxyl group, C2-4 polyhydroxyalkoxyl group, or 2,4-diaminophenoxyalkoxyl group.

 $\begin{tabular}{ll} \hline [0050] & Typical examples of the m-phenylenediamine represented by the general formula (6) above are listed below: 2,4-diaminobenzene, 3,5-diamino-1-ethyl-2-methoxybenzene, 3,5-diamino-2-methoxy-1-methylbenzene, 2,4-diaminobenzene, 1,3-bis(2,4-diaminophenoxy)propane, bis(2,4-diaminophenoxy)methane, 1-(<math>\beta$ -aminoethyloxy)-2,4-diaminobenzene, 2-amino-1-(β -hydroxyethyloxy)-4-methylaminobenzene, 2,4-diamino-1-ethoxy-5-methylbenzene, 2,4-diamino-5-(β -hydroxyethyloxy)-1-methylbenzene, 2,4-diamino-1-(β -hydroxyethyloxy)-1-methylbenzene, 2,4-diamino-1-(β -hydroxyethyloxy)benzene, 2-amino-4-N-(β -hydroxyethyl)-amino-1-methoxybenzene, and acid salts thereof.

[0051] The m-diphenyl used as the coupler in the composition of the present invention should preferably be any compound or acid salt thereof represented by the general formula (7) below.

$$\begin{array}{c}
OH \\
R^{26} \\
OH
\end{array}$$
(7)

[0052] R²⁶ and R²⁷ (which may be identical or different) each denotes a hydrogen atom, C1-4 alkyl group, or halogen atom which is any of chlorine, iodine, bromine, and fluorine.

[0053] Typical examples of the m-diphenol represented by the general formula (7) above are: 1,3-dihydroxybenzene, 2-methyl-1,3-dihydroxybenzene, 4-chloro-1,3-dihydroxybenzene, 2-chloro-1,3-dihydroxybenzene, and acid salts thereof.

[0054] The heterocyclic coupler used as the coupler in the composition of the present invention includes the following: Benzimidazole derivatives, benzmorpholine derivatives, sesamol derivatives, pyrazolo-azole derivatives, pyrolo-azole derivatives, imidazole-azole derivatives, pyrazolopyrlmidine derivatives, pyrazolin-3,5-dion derivatives, pyrolo-[3,2d]-oxazole derivatives, pyrazol[3,4d]thiazole derivatives, thiazolo-azole S-oxide derivatives, thiazolo-azole S,S-dioxide derivative, and acid salts thereof.

[0055] The heterocyclic coupler that can be used as the coupler in the composition of the present invention includes the following:

Benzimidazole derivatives, benzmorpholine derivatives, sesamol derivatives, pyrazolo-azole derivatives, pyrrolo-azole derivatives, imidazolo-azole derivatives, pyrazolopyrimidine derivatives, pyrazolin-3,5-dion derivatives, pyrrolo-[3,2d]-oxazole derivatives, pyrazolo[3,4d]-thiazole derivatives, thiazolo-azole S-oxide derivatives, thiazolo-azole S,S-dioxide derivatives, and acid salts thereof.

[0056] The benzimidazole derivative as the heterocyclic coupler that can be added to the composition of the present

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invention includes any compound and acid salts thereof represented by the general formula (8) below.

$$R^{30}$$
 R^{31}
 R^{29}
 R^{29}
 R^{29}
 R^{29}

[0057] R¹⁸ denotes a hydrogen atom or C1-4 alkyl group; R²⁹ denotes a hydrogen atom, C1-4 alkyl group, or phenyl group; and R³⁰ denotes a hydrogen atom, methoxy group, or C1-4 alkyl group. If R³⁰ is an amino group, it is at the fourth position; if R³⁰ is at the fourth position, R³¹ is at the seventh position; and if R³⁰ is at the fifth position, R³¹ is at the sixth position.

[0058] Typical examples of the benzimidazole derivative represented by the general formula (8) above are listed below:

4-hydroxybenzimidazole, 4-aminobenzimidazole, 4-hydroxy-7-methylbenzimidazole, 4-hydroxy-2-methylbenzimidazole, 1-butyl-4-hydroxybenzimidazole, 4-amino-2-methylbenzimidazole, 5,6-dihydroxybenzimidazole, 5-hydroxy-6-methoxybenzimidazole, 4,7-dihydroxy-1-methylbenzimidazole, 4,7-dimethoxybenzimidazole, 5,6-dihydroxy-1-methylbenzimidazole, 5,6-dihydroxy-2-methylbenzimidazole, 5,6-dimethoxybenzimidazole, and acid salts thereof.

[0059] The benzomorpholine derivative as the heterocyclic coupler that can be added to the composition of the present invention includes any compound and acid salts thereof represented by the general formula (9) below.

$$\begin{array}{c|c}
C & & (9) \\
\hline
Z & & \\
R^{32} & &
\end{array}$$

[0060] R³² and R³³ (which may be identical or different) each denotes a hydrogen atom, C1-4 alkyl group, C1-4 monohydroxyalkyl group, C2-4 polyhydroxyalkyl group, (C1-4)alkoxy(C1-4)alkyl group, nitrogen-containing group, or phenyl group; and Z denotes a hydroxyl group or amino group.

[0061] Typical examples of the benzomorpholine derivative represented by the general formula (9) above include 6-hydroxy 1,4-benzomorpholine, N-methyl 6-hydroxy-1,4-benzomorpholine, 6-amino 1,4-benzomorpholine, and acid salts thereof.

[0062] The sesamol derivative as the heterocyclic coupler that can be added to the composition of the present invention includes any compound and acid salts thereof represented by the general formula (10) below.

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[0063] R³⁴ denotes a hydroxyl group, amino group, C1-4 alkylamino group, C1-4 monohydroxyalkylamino group, or C2-4 polyhydroxyalkylamino group; and R³⁵ denotes a hydrogen atom, halogen atom, or C1-4 alkoxyl group.

[0064] Typical examples of the sesamol derivative represented by the general formula (10) above include 2-bromo 4,5-methyleneoxyaniline, 2-(β-hydroxyethyl)amino 4,5-methylenedioxybenzene, and acid salts thereof.

[0065] The pyrazole-azole derivative as the heterocyclic coupler that can be added to the composition of the present invention includes those compounds mentioned in the following patents and literatures:

Patents: FR-2075583, EP-A-119860, EP-A-285274, EP-A-244160, EP-A-578248, GB 1458377, US 3277554, US 3419391, US 3061432, US 4500630, US 3725067, US 3926631, US 5457210, JP 84/99437, JP 83/42045, JP 84/162548, JP 84/171956, JP 85/33552, JP 85/43659, JP 85/172982, and JP 85/190779.

Literatures: Chem. Ber. 32, 797 (1899), Chem. Ber. 89, 2550, (1956), J. Chem. Soc. Perkin Trans., 2047, (1977), J. Prakt. Chem., 320, 533, (1978).

[0066] Typical examples of the above-mentioned pyrazolo-azole derivative are listed below:

2-methylpyrazolo [1,5-b]-1,2,4-triazole, 2-ethylpyrazolo [1,5-b]-1,2,4-triazole, 2-isopropylpyrazolo [1,5-b]-1,2,4-triazole, 2-phenylpyrazolo [1,5-b]-1,2,4-triazole, 2,6-dimethylpyrazolo [1,5-b]-1,2,4-triazole, 7-chloro 2,6-dimethylpyrazolo [1,5-b]-1,2,4-triazole, 3,6-dimethylpyrazolo [3,2-c]-1,2,4-triazole, 6-phenyl-3-methylthiopyrazolo [3,2-c]-1,2,4-triazole, 6-aminopyrazolo [1,5,a]-benzimidazole, and acid derivative thereof.

[0067] The pyrrolo-azole derivative as the heterocyclic coupler that can be added to the composition of the present invention includes those compounds mentioned in the following patents and literatures.

Patents: US 5 256 526, EP-A-557851, EP-A-577248, EP-A-578248, EP-A-518238, EP-A-456226, EP-A-488909, EP-A-488248

Literatures: D. R. Liljegren Ber. 1964, 3436; E. J. Browne, J. C. S., 1962, 5149; P. Magnus, J. A. C. S., 1990, 112, 2465; P. Magnus, J. A. C. S., 1987, 109, 2711; Angew. Chem. 1960, 72, 956, and Rec. Trav. Chim. 1961, 80, 1075.

[0068] Typical examples of the above-mentioned pyrrolo-azole derivative include 5-cyano-4-ethoxycarbonyl-8-methylpyrrolo [1,2-b]-1,2,4-triazole, 5-cyano-8-methyl-4-phenylpyrrolo [1,2-b]-1,2,4-triazole, 7-amino-6-ethoxycarbonylpyrrolo [1,2-a]-benzimidazole, and acid derivatives thereof.

[0069] The imidazolo-azole derivative as the heterocyclic coupler that can be added to the composition of the present invention includes those compounds mentioned in US 5441863, JP 62-279337, JP 6-236011, and JP 7-92632.

[0070] Typical examples of the above-mentioned imidazolo-azole derivative include 7,8-diaminoimidazolo-[3,2-a]-imidazole, 7,8-dicyano-4-methylimidazolo-[3,2-a]-imidazole, and acid salts thereof.

[0071] The pyrazolo-pyrimidine derivative as the heterocyclic coupler that can be added to the composition of the present invention includes those compounds mentioned in EP-A-304001.

[0072] Typical examples of the above-mentioned pyrazolopyrimidine derivative include the following. Pyrazolo-[1,5-a]pyrimidin-7-on, 2-methyl-6-ethoxycarbonylpyrazolo-[1,5-a]pyrimidin-7-on, 2-methyl-5-methoxymethylpyrazolo-[1,5-a]pyrimidin-7-on, 2-tert-butyl-5-fluoromethylpyrazolo-[1,5-a]pyrimidin-7-on, 2,7-dimethylpyrazolo-[1,5-a]pyrimidin-5-on, and acid derivatives thereof.

[0073] The pyrazolin-3,5-dione derivative as the heterocyclic coupler that can be added to the composition of the present invention includes those compounds mentioned in the following patents and literatures. Patents: JP 7-36159, JP 7-84348, and US 4128425.

Literatures:

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- (1) L. WYZGOWSKA, Acta. Pol. Pharm. 1982, 39(1-3), 83
- (2) E. HANNING, Pharmazie, 1980, 35(4), 231
- (3) M. H. ELNAGDI, BULL. Chem. Soc. Jap., 46(6), 1830, 1973
- (4) G. CARDILLO, Gazz. Chim. Ital. 1966, 96, (8-9), 973.

[0074] Typical examples of the above-mentioned pyrazolin-3,5-dione derivative include 1,2-diphenylpyrazolin-3,5-dione, 1,2-diethylpyrazolin-3,5-dione, and acid salts thereof.

[0075] The pyrrolo-[3,2-d]-oxazole derivative as the heterocyclic coupler that can be added to the composition of the present invention includes those compounds mentioned in JP 7-325375 and J. Heterocycl. Chem. 16, 13, (1979).

[0076] The pyrazolo-[3,4-d]-thiazole derivative as the heterocyclic coupler that can be added to the composition of the present invention includes the compound mentioned in Japanese Patent Laid-open No. Hei 7-244361.

[0077] The thiazolo-azole S-oxide derivative and thiazolo-azole S,S-dioxide derivative as the heterocyclic coupler that can be added to the composition of the present invention include those compounds mentioned in the following patents and literatures:

- (1) JP 7-98489
- (2) Khim. Geterotsilk. Sodein, 1967, p. 93.

- (3) J. Prakt. Chem., 318, 1976, p. 12.
- (4) Indian J. Heterocycl. Chem. 1995, 5(2), 135.
- (5) Acta. Pol. Pharm. 1995, 52(5), 415.
- (6) Heterocycl. Commun. 1995, 1(4), 297.
- (7) Arch. Pharm. (Weinheim, Ger.), 1994, 327(12), 825.

[0078] The above-mentioned coupler should be added in an amount of 0.0001 to 10%, preferably 0.005 to 5%, of the total amount of the composition of the present invention.

[0079] The composition of the present invention may optionally be incorporated with a cationic direct dye. The cationic direct dye that can be added to the composition of the present invention includes, for example, cationized aminoanthraquione dyes, cationized mono- or di-azo dyes, and cationized naphthoquinone dyes.

[0080] Typical examples of the above-mentioned dyes are listed below:

[8-[(p-aminphenyl)azo]-7-hydroxy-2-naphthyl]trimethylammonium chloride (synonymous with basic brown 16, arianol mahogany 306002 in color index), 3-[(4-amino-6-bromo-5,8-dihydro-1-hydroxy-8-imino-5-oxo-2-naphthalenyl)amino] -N,N,N-trimethylbenzenamium chloride (synonymous with basic blue 99, arianol steel blue 306004 in color index), 7-hydroxy-8-[(2-methoxyophenyl)azo]-N,N,N-trimethyl-2-naphthalenamium chloride (synonymous with basic red 76, arianol madar red in color index), [8-(4-amino-2-nitrophenyl)azo]-7-hydroxy-2-naphthyl]trimethyl ammonium chloride (cynonymous with basic brown 17, arianol siena brown 306001 in color index), 3-[(4,5-dihydro-3-methyl-5-oxo-1-phenyl-1H-pyrazol-4-yl)azo]-N,N,N-trimethylbenzenamium chloride (synonymous with basic yellow 57, arianol straw yellow 306005 in color index).

[0081] In addition, the above-mentioned cationic direct dye may be selected from the following.

(a) Those compounds represented by the general formula (11) below.

$$\begin{array}{c|c}
R^{38} \\
\hline
R^{36} \\
\hline
R^{37}
\end{array}$$
(11)

D denotes a nitrogen atom or -CH group; R^{36} and R^{37} (which may be identical or different) each denotes a hydrogen atom, C1-4 alkyl group (which may be substituted with any one of -CN, -OH, and -NH₂, or accompany a carbon atom and form a benzene ring or an oxygen-containing or nitrogen-containing hetero ring (with the ring optionally being substituted with one or more C1-4 alkyl group or 4'-aminopenyl group).

R³⁸ and R³⁸ (which may be identical or different) each denotes a hydrogen atom, halogen atom selected from chlorine, bromine, iodine, and fluorine, cyano group, C1-4 alkoxyl group, or acetyloxy group; X⁻ denotes an anion (preferably chloride, methylsulfate, or acetate); and A denotes any of the following groups numbered A1 to A19.)

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$$R^{39}$$
 R^{39} R^{39}

 R^{39} denotes a C1-4 alkyl group (which may be substituted with a hydroxyl group or C1-4 alkoxyl group); and R^{39} denotes a C1-4 alkoxyl group.

(b) Those compounds represented by the general formula (12) below.

$$\begin{array}{c|c}
R^{42} \\
R^{40} \\
R^{41}
\end{array}$$

$$\begin{array}{c|c}
R^{40} \\
R^{41}
\end{array}$$

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R⁴⁰ denotes a hydrogen atom, C1-4 alkyl group; R⁴¹ denotes a hydrogen atom or alkyl group (which may be substituted with any one of -CN group, amino group, 4'-aminophenyl group) or R⁴¹ may form an oxygen-containing and/or nitrogen-containing hetero aromatic ring with R⁴⁰ (this hetero aromatic ring may be substituted with a C1-4 alkyl group); R⁴² and R⁴³ (which may be identical or different) each denotes a hydrogen atom, halogen atom selected from chlorine, bromine, iodine, and fluorine, C1-4 alkoxyl group, C1-4 alkoxyl group, or -CN group; X-denotes an anion (preferably chloride, methylsulfate, or acetate); and B denotes any one of the following groups numbered B1 to B6.

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B1 B2
$$\mathbb{R}^{46}$$
 B3 \mathbb{R}^{44} ; \mathbb{R}^{44} \mathbb{R}^{44} \mathbb{R}^{44} \mathbb{R}^{44} \mathbb{R}^{44} \mathbb{R}^{44} \mathbb{R}^{44} \mathbb{R}^{5} \mathbb{R}^{6}

R⁴⁴ denotes a C1-4 alkyl group, and R⁴⁵ and R⁴⁶ (which may be identical or different) each denotes a hydrogen atom or C1-4 alkyl group.

(c) Those compounds represented by the general formulas (13) and (13') below.

$$E-D_{1} = D_{2} - (N)_{\overline{m}} - R^{47} - X$$

$$R^{51} = D_{1} = D_{2}$$

$$R^{51} = D_{1} = D_{2}$$

$$R^{51} = D_{1} = D_{2}$$

$$R^{50} = D_{1} = D_{2}$$

$$R^{51} = D_{2}$$

$$R^{51} = D_{1} = D_{2}$$

$$R^{51} = D_{2}$$

$$R^$$

[0082] R^{47} denotes a hydrogen atom, halogen atom selected from chlorine, bromine, iodine, and fluorine, amino group, C1-4 alkoxyl group, or acetyloxy group; R^{48} denotes a hydrogen atom, C1-4 alkyl group, or heterocyclic ring containing carbon atoms (forming a benzene ring in the molecule) or oxygen atom (this hetero ring may be substituted with a C1-4 alkyl group); R^{49} denotes a halogen atom selected from chlorine, bromine, iodine, and fluorine; R^{50} and R^{51} (which may be identical or different) each denotes a hydrogen atom or C1-4 alkyl group; D_1 and D_2 (which may be identical or different) each denotes a nitrogen atom or -CH group; m is 0 or 1. If R^{47} is an unsubstituted amino group, both D_1 and D_2 denote a -CH group, and m is 0. X- denotes an anion, preferably chloride, methylsulfate, or acetate. E is any one of the following groups numbered E1 to E8.

[0083] R⁵² denotes a C1-4 alkyl group.

[0084] If m is 0 and D₁ denotes a nitrogen atom, E denotes E9 given below.

E6

[0085] R52 denotes a C1-4 alkyl group.

[0086] The compounds represented by the general formulas (11), (12), (13), and (13') above as the cationic direct dye that can be added to the composition of the present invention include, for example, those compounds mentioned in WO 95/01772, WO 95/15144, and EP-A-0714954.

E7

E8

[0087] Typical examples of the compounds represented by the general formula (11) as the cationic direct dye that

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can be added to the composition of the present invention include those which are represented by the following structural formulas (11-1) to (11-52).

$$CH_3$$
 $N = N$
 $N = N$
 CH_3
 CI
 CH_3
 CI
 CH_3

$$CH_3$$
 N
 N
 N
 N
 CH_3
 CH_3
 CH_3

$$H_3C-N$$
 $CH=CH-CH$
 CH_3
 CH_3
 CH_3

$$CH_3$$
 CH_3 CH_3 CH_3 CH_3 CH_3 CH_3 CH_3

$$H_3C-N_+$$
 $CH=CH-CH_3$ CI (11-5)

$$_{55}$$
 HO- $_{H_4}C_2$ - $_{N_+}$ CH- $_{CH_3}$ CH- $_{CH_3}$ CI (11-6)

$$H_3C-N+$$
 $CH=CH CH_3$
 CI (11-7)

$$N+N+N=N-N+2$$
 Cl (11-10)

CH₃

$$N+ N = N - NH_2 \quad CI$$

$$CH_3 \quad OCH_3$$

$$CH_4 \quad OCH_3$$

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$$CH_3$$
 $N+$
 $N=N C_2H_5$
 C_2H_5
 C_2H_5
 C_2H_5

$$CH_3$$
 $N+$
 $N=N C_2H_4-CN$
 C_2H_4-CN
 CH_3

$$N^+$$
 N^+
 CH_3
 CH_3
 CH_3
 CI
 NH_2
 CI
 $(11-15)$

$$CH_3$$
 $N+$
 $N=N$
 NH_2
 CH_3
 CH_3
 CH_3

$$CH_3$$
 N
 N
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3

Cl ·

(11-27)

$$N+$$
 $N=N O-CH_3$
 $O-CH_3$
 $O-CH_3$
 $O-CH_3$
 $O-CH_3$
 $O-CH_3$
 $O-CH_3$

$$CH_3$$
 $O-CH_3$ CH_3 CH_3 CH_3 CH_3 CH_3 CH_3 CH_3 CH_3

$$H_3C-N+$$
 $N=N CH_3$
 CH_3
 CH_3

$$CH_3$$
 $N = N$
 NH_2
 CH_3
 CH_3

$$N = N$$
 $N = N$
 CH_3
 CH_3
 $N = N$
 CH_3
 $N = N$
 $N = N$

$$CH_3 - N_+$$
 $N = N$
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3

 H_3C-O N=N+ N=N+

$$N = N$$
 CH_3
 CH_3
 CI
 CI
 CI
 CI
 CI

$$N$$
 N
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3

$$CH_3$$
 $N=N_+$
 $N=N_+$
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3

$$H_3C$$
 $N+$
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3

$$\begin{array}{c|c}
N & CH_{3} \\
N+ & CH_{3}
\end{array}$$

$$\begin{array}{c}
CH_{3} \\
CH_{4}
\end{array}$$

$$\begin{array}{c}
CH_{3} \\
CH_{3}
\end{array}$$

$$CH_3$$
 N^+
 $N=N$
 CH_3
 CH_3
 CH_3
 CH_3

$$CH_3$$
 N^+
 $N=N$
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3

$$N^+$$
 $N=N$
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3

$$CH_3$$
 N^+
 $N=N$
 CH_3
 CH_3
 CH_3
 CH_3

$$CH_3$$
 CH_3
 CH_3
 CH_3SO_4
 CH_3SO_4
 CH_3SO_4
 CH_3SO_4
 CH_3SO_4

$$V_{2}H_{5}$$
 $V_{2}H_{5}$
 V_{3}
 V_{4}
 V_{5}
 V_{5}
 V_{6}
 V_{7}
 $V_$

$$CH_3$$
 N^+
 $N=N$
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3

$$CH_3$$
 N
 N
 CH_3
 CH_3

35 [0088] Of the compounds represented by the structural formulas (11-1) to (11-52) above, those which are represented by the structural formulas (11-1), (11-2), (11-4), (11-14), and (11-31) are particularly preferable.

[0089] Those compounds represented by the general formula (12) above as the cationic direct dye that can be added to the composition of the present invention include, for example, those compounds which are represented by the structural formulas (12-1) to (12-12) below. Of these compounds, those which are represented by the structural formulas (12-1) and (12-12) are particularly preferable.

$$H_3C$$
 $N=N$
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3

$$H_3C$$
 N
 N
 N
 N
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3

$$H_3C$$
 $N+$
 $N=N N$
 CH_3
 CH_3SO_4 (12-6)
 CH_3

$$CH_3$$
 $N+$
 $N=N$
 CH_2 - CH_2 - CN
 Cl^- (12-9)
 CH_3

$$H_3C$$
 N
 CH_3
 CH_3
 CH_3SO_4
 CH_3SO_4
 CH_3SO_4
 CH_3SO_4

[0090] Those compounds represented by the general formula (13) above as the cationic direct dye that can be added to the composition of the present invention include, for example, those compounds which are represented by the structural formulas (13-1) to (13-18) below.

$$\begin{array}{c|c}
S \\
CH = N - N \\
CH_3
\end{array}$$

$$Cl \cdot (13-1)$$
;

$$H_3C$$
 $CH=N-N$
 $CH=1$
 $CH=1$
 $CH=1$
 $CH=1$
 $CH=1$
 $CH=1$

$$H_3C$$
 $CH = N$
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3

$$H_3C-N+$$
 $CH=N-N CH_3SO_4$
 CH_3SO_4
 CH_3SO_4
 CH_3SO_4

$$H_3C-N+$$
 $CH=N-N OCH_3$ Cl^- (13-5)

$$H_3C-N_+$$
 $CH=N-N_ OCH_3$ CH_3SO_4 (13-13)

$$CH$$
 CH CH_3 CH_3COO (13-15)

$$H_3C-N_1$$
 $CH=N-N_1$
 CH_3
 CI (13-17)

;

$$CI$$
 $N=N$
 $N+$
 CI
 CH_2
 CI
 $(13-18)$

[0091] Of the compounds represented by the structural formulas (13-1) to (13-18) above, those which are represented by the structural formulas (13-4), (13-5), and (13-13) are particularly preferable.

[0092] Those compounds represented by the general formula (13') above as the cationic direct dye that can be added to the composition of the present invention include, for example, those compounds which are represented by the structural formulas (13'-1) to (13'-3) below.

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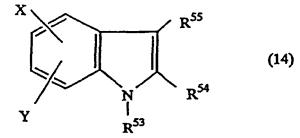
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$$CH_3$$
 $N+$
 CH_3
 CH_3
 CH_3
 CH_3

[0093] The above-mentioned cationic direct dye should be added in an amount of 0.001 to 10%, preferably 0.05 to 5%, of the total amount of the composition of the present invention.

[0094] In general, the acid salts (oxidative base compounds and couplers) which are preferable from the standpoint of the composition of the present invention are hydrochloride, hydrobromide, sulfate, succinate, lactate, and acetate. [0095] A melanin precursor-like substance represented by the formula (14) below is also preferable from the recent nature-loving view point.



[0096] X denotes a hydrogen atom, NH₂, OH, C1-6 linear or branched alkyl group, alkenyl group, or alkoxyl group; 55 and Y denotes a hydrogen atom, OH, or NH2. If X denotes OH or C1-6 linear or branched alkyl group, alkenyl group, or alkoxyl group, X is at the position 5, 6, or 7 of the ring, and at the ortho position with respect to Y.

[0097] Also, R53 and R55 (which may be identical or different) each denotes a hydrogen atom or C1-6 linear or

branched alkyl group, alkenyl group, or alkoxyl group; R⁵⁴ denotes a hydrogen or C1-6 linear or branched alkyl group, alkenyl group, alkoxyl group, or carboxyl group.

[0098] Typical examples of the compound represented by the general formula (14) above are listed below: 4,5-dihydroxyindole, 5,6-dihydroxyindole, 6,7-dihydroxyindole, N-methyl-5,6-dihydroxyindole, N-methyl-5,6-dihydroxyindole, N-methyl-5,6-dihydroxyindole, N-methyl-5,6-dihydroxyindole, 2,3-dimethyl-5,6-dihydroxyindole, 2-methyl-5-ethyl-6-hydroxyindole, 2-methyl-5-hydroxy-6-β-hydroxyethylindole, 4-hydroxypropylindole, 2,3-dimethyl-5,6-dihydroxyindole, 4-hydroxy-5-methoxyindole, 6-hydroxy-7-methoxyindole, 6-hydroxy-5-methoxyindole, 6-hydroxyindole, 5-hydroxyindole, 7-hydroxyindole, 7-aminoindole, 5-aminoindole, 4-aminoindole, 5,6-dihydroxyindole carboxylic acid, 1-methyl-5,6-dihydroxyindole, and salts thereof.

[0099] It is also desirable to use the melanin precursor-like substance represented by the general formula (15) below.

$$\begin{array}{c|c}
R^{58} \\
\hline
R^{56}
\end{array}$$
(15)

[0100] K denotes a hydrogen atom, NH₂, OH, C1-6 linear or branched alkyl group, alkenyl group, or alkoxyl group; and L denotes OH or NH₂. If K denotes OH or C1-6 linear or branched alkyl group, alkenyl group, or alkoxyl group, K is at the fifth, sixth, or seventh positions of the ring and at the ortho position with respect to L.

[0101] Also, R⁵⁶ and R⁵⁸ (which may be identical or different) each denotes a hydrogen atom or C1-6 linear or branched alkyl group, alkenyl group, or alkoxyl group; R⁵⁷ denotes a hydrogen or C1-6 linear or branched alkyl group, alkenyl group, or carboxyl group.

[0102] Typical examples of the compound represented by the general formula above are listed below: 4,5-dihydroxyindoline, 5,6-dihydroxyindoline, N-methyl-5,6-dihydroxyindoline, N-methyl-5,6-dihydroxyindoline, N-methyl-5,6-dihydroxyindoline, N-methyl-5,6-dihydroxyindoline, N-methyl-5,6-dihydroxyindoline, N-methyl-5,6-dihydroxyindoline, N-methyl-5,6-dihydroxyindoline, 2-methyl-5,6-dihydroxyindoline, 2-methyl-5-hydroxyindoline, 2-methyl-5-hydroxy-6-phydroxyethylindoline, 4-hydroxypropylindoline, 2,3-dimethyl-5,6-dihydroxyindoline, 4-hydroxy-5-methoxyindoline, 6-hydroxy-7-methylindoline, 6-hydroxy-5-methoxyindoline, 6-hydroxyindoline, 5-hydroxyindoline, 7-hyroxyindoline, 7-methyl-5,6-dihydroxyindoline, 5-methyl-5,6-dihydroxyindoline, 3-methyl-5,6-dihydroxyindoline, 3-methyl-5,6-dihydroxyindo

[0103] The above-mentioned oxidative color-developing substances may be used alone or in combination with one another for adequate color matching.

[0104] The amount of the above-mentioned oxidative color-developing substance in the composition for dyeing keratinous fiber according to the present invention is not specifically restricted. It may be properly increased or reduced for color matching depending on the type of the commodity of the composition and the kind of the oxidative color-developing substance. Usually, it is added in an amount of 0.01 to 20%, preferably 0.1 to 10%, of the total amount of the composition. If the amount of the oxidative color-developing substance is excessive, the amount of the oxidase necessarily decreases to such an extent that it does not produce the oxidizing action, and it is hard to stabilize it. Incidentally, if the amount of the oxidative color-developing substance is too small, the desired effect may not be produced.

[0105] The oxidase of the present invention is an oxidase which acts on oxygen as the substrate but does not evolve hydrogen peroxide as mentioned above. A four-electron reductive oxidase is known as such an oxidase, which includes, for example, catechol oxidase, amine oxidase, and laccase. They may be used alone or in combination with one another. [0106] In the composition of the present invention, the amount of the above-mentioned enzyme is not specifically restricted. It varies depending on the type of the product, frequency of use, treatment time, and the potency of the enzyme. For example, it may be used in an amount of 0.01 to 50%, preferably 0.1 to 30%, of the total amount of the composition. If the amount is too small, the above-mentioned oxidase does not fully produce its effect. If the amount is too large, the effect of the oxidase does not increase in proportion to the added amount.

[0107] According to the present invention, the amount of the enzyme should preferably be in the above-mentioned range. In the case where the amount of the enzyme is specified, it is desirable to specify the amount on the basis of the active value of the enzyme. It is useful to use the measured value of the amount of dissolved oxygen in the reaction

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system. The enzyme reaction is a reaction which consumes oxygen in the reaction system and polymerizes the color precursor. It is possible to measure the enzyme activity by monitoring the amount of oxygen consumed.

[0108] A commonly used method for determining the amount of dissolved oxygen is the one which uses an oxygen electrode. This method is simple, highly reproducible, and comparatively accurate. In the case of an enzyme reaction system involving a simple combination of oxygen, color-developing oxidative substance, and enzyme, it used to be difficult to obtain constant measured values by measurement with an oxygen electrode; however, it is possible to obtain stable measured values by controlling the temperature for measurement at 0 to 70°C and by using an adequate buffer solution to adjust the pH value to 6 to 8. In the case where it exists in the form of composition, it is possible to obtain stable measured values by setting up the temperature condition and pH condition in the same way as above although the measurement scale is larger than that in the simple system.

[0109] The oxygen electrode as the measuring apparatus is available in three types: oxygen balance type probe, galvanic type probe, and polaragraphic type probe. Any of them gives require measured values when used for this object. The enzyme as the active ingredient in the composition to be determined can be determined without restriction of its origin. It should preferably be a four-electron reductive oxidase. Its typical examples include laccase, polyphenol oxidase, and glucose oxidase. To be more specific, they are laccase (E.C. 1.10.3.2), catechol oxidase (E.C. 1.10.3.1), bilirubin oxidase (E.C. 1.3.3.5), and monophenol monooxidase (E.C. 1.14.99.1). Laccase is an enzyme containing a plurality of copper atoms which catalyzes the oxidation of phenols or aromatic amine compounds. The oxidizing reaction by laccase gives rise to aryloxy radials from an adequate phenolic compound. This reaction product yields a dimmer, oligomer, or polymer by polymerization reaction. This laccase originates from microorganisms (such as fungi and bacteria) or plants. Those originating from fungi are preferable. To be more specific, those originating from the following fungi or plants are preferable.

Fungi:

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Polyporus sp. (e.g., P. pinsitus and P. versicolor)
Mycerliophthora sp. (e.g., M. thermophila)
Phizocutonia sp. (e.g., Rh. praticola and Rh. solani)
Pyriculania sp. (e.g., P. oryzae)
Scytalidium sp (e.g., S. thermophilium)

30 Plants:

Rhus sp. (e.g., Rhus vernicifera)

[0110] Among oxidation-reduction enzymes are known the following:

Laccase originating from Polyporus sp. (specifically Polyporus pinisitus-originating laccase) which laccase is also called Trametes Villos-originating laccase disclosed in WO 96/00290 (NOVO Nordisk Biotec Inc.), Laccase originating from Mytheliophthora thermophila disclosed in WO 95/33836 (NOVO Nordisk Biotec Inc.), Laccase originating from Scytalidium sp. (specifically S. thermophilium-originating laccase) disclosed in WO 95/33837 (NOVO Nordisk Biotec Inc.). Included in them are laccase originating from Pyriculania sp. (Pyricularia oryzae) which is commercially available from SIGMA Corp. under a trade name of L5510, laccase originating from Coprinus sp. (C. Cinereus), and laccase originating from Rhizoctonia sp. (Rh. solani) with an optimum pH 6.0-8.5 disclosed in WO 95/07988.

[0111] Other known laccases are those originating from the following fungi: Collybia, Fomes, Letium, Pleurotus, Aspergillus, Neutospora, Podospora, Phlebia (P. radiata) disclosed in WO 92/0104, Coriolus sp. (C. hirsitus) disclosed in JP 2-238885, and Botrytis.

[0112] A preferred bilirubin oxidase is the one originating from Mycrothecim sp. (M. verrucaria).

[0113] The H_2O_2 -producing oxidase is usually used in combination with a peroxide which decomposes H_2O or decreases the production of H_2O_2 . Such a peroxidase includes, for example, glucose oxidase (E.C. 1.1.3.4), hexose oxidase (E.C. 1.1.3.5), L-amino acid oxidase (E.C. 1.4.3.2), xylitol oxidase, galactose oxidase (E.C. 1.1.3.9), pyranose oxidase (E.C. 1.1.3.10), and alcohol oxidase (E.C. 1.1.3.13).

[0114] The L-amino acid oxidase should preferably be the one which originates from Trichloderma sp. (specifically T. harzianum) disclosed in WO 94/25574 NOVO Nordisk A/S, or the one which originates from T. viride.

[0115] The glucose oxidase should preferably be one which originates from Aspergillus sp. (A. niger) or Cladosporium sp. (specifically C. oxysprorum).

[0116] The hexose oxidase is an enzyme which oxidizes carbohydrates such as D-glucose. D-galactose, maltose, cellobiose, lactose, D-glucose-6-phosphate, D-mannose, 2-deoxy-D-glucose, 2-doxy-D-galactose, D-fructose, D-glucuronic acid, and D-xylose, which originate from Chondrus. Crispus (known as Irish moss) as red algae. (Sullivan and Ikawa, (1973), Biochim. Biophys. Acts, 309, p. 11-22; Ikawa, (1982), Meth. in Enzymol. 89, Carbohydrate Metabolism Part D, 145-149)

[0117] The measuring method mentioned herein can be applied to them regardless of their origin. In the composition of the present invention, the amount of the oxidase should be 0.005-10.0, preferably 0.01-5.0, in terms of active ingredient (Δ DO value) specified by the measuring method. This amount is adequate for good dyeing property. The oxidase regardless of its origin can be used in the composition of the present invention. More than one oxidase may be used in combination with one another.

[0118] The measurement of activity employs as the substrate color-developing substances such as dye precursor, developer, and direct dye. Their kind and amount are not specified for color matching. Ordinary oxidation dyes mentioned above can be used, which include the one mentioned in the standard for hair dye raw materials (The 3rd revised edition, issued in May 1985, by Japan Hair Color Industry Association, Hair Dye Forum).

[0119] The practical method for determination is explained in the following: .

(1) Method for determination

(Apparatus)

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[0120] Dissolved oxygen meter, thermostat, incubator (102 mL), beaker (made of glass), magnetic stirrer, stirrer bar, clamp, timer, and measuring flask.

(Substrates for measurment)

[0121]

	. (6)										5.5	0.2	2	-	0.2	0.2	Laccase	-	,	9	Balance
5	(i) .	0.5	-	0.5	0.05	0.15			2		1.5	0.2	2	-	0.2	0.2	Uricase			9	Balance
10	(h)	0.5	-	0.5	0.05	0.15			2		1.5	0.2	2	1	. 0.2	⊕ 0.2	Laccase		·	10	Balance
15	(6)	0.5		0.5	0.5		0.05	0.1	2		0.05	-	2.5	1.5	0.15	0.3	Laccase			5	Balance
20	(1)	0.5	-	0.5					2	0.15	9.0	2	2	-	0.2	0.2	Laccase			10	Balance
	*(e)																Laccase	0.02	Balance		
25	*(b)	-	0.2	0.2	0.4	0.04											Uricase		Balance	96	
30	(0)	8.0			9.0	0.4	-										Laccase			96	Balance
35	(q)	0.2	4.0	-													Laccase			96	Balance
40	(a)	2															Laccase			96	Balance
45										itoluene									xide buffer (pH		
50			Je	ine					e sulfate	amino)-5-amino	ər			nulsifying agent		ary			ootassium hydro		
55		p-aminophenol	p-phenylenediamine	m-phenylenediamine	m-aminophenol	m-diphenol	o-aminocresol	Resorcin	2,5-diaminotoluene sulfate	2-(2'-hydroxyethylamino)-5-aminotoluene sulfate	Polymeric thickener	pH adjustor	Antioxidant	Dispersing and emulsifying agent	liO	Permeation auxiliary	Enzyme	Uric acid	50 mM boric acid-potassium hydroxide buffer (pH 8.5)	Ethanol	Purified water

Asterisk (*) indicates comparison. Total amount is 100 mass%.

(Method 1)

[0122]

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- (1) Place the buffer solution for measurement in the incubator or beaker, and keep it warm.
- (2) Add the substrate for measurement to the buffer solution, and uniformly mix the solution with the stirrer.
- (3) Insert the oxygen electrode and mix the solution with the stirrer to stabilize the system.
- (4) Add the enzyme sample and start the timer simultaneously.
- (5) Determine the amount of dissolved oxygen at a certain interval, and record the difference (ΔDO) of two measurements.

(Method 2)

[0123]

[012.

- (1) Place the buffer solution for measurement in the incubator or beaker, and keep it warm.
- (2) Add the enzyme sample to the buffer solution, and uniformly mix the solution with the stirrer.
- (3) Insert the oxygen electrode and mix the solution with the stirrer to stabilize the system.
- (4) Add the substrate for measurement and start the timer simultaneously.
- (5) Determine the amount of dissolved oxygen at a certain interval, and record the difference (ΔDO) of two measurements.

(Method 3)

25 [0124]

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- (1) Place a predetermined amount the composition sample (containing the enzyme) in the incubator or beaker.
- (2) Add the buffer solution (previously warmed) and run the stirrer.
- (3) Insert the oxygen electrode and start the timer.
- (4) Determine the amount of dissolved oxygen at a certain interval, and record the difference (ΔDO) of two measurements.

(Results)

³⁵ [0125] The method 1 or 2 was used for the substrates (a) to (e), the method 3 was used for the substrates (f) to (j). The results of determination are shown in Table 1 below.

[0126] It is noted that the value of Δ DO changes as the amount of the enzyme added (or in the composition) increases. This suggests that the above-mentioned method is suitable for determining the amount of the enzyme as the active ingredient in the composition. It is also noted from the results of comparative example that uric acid is inadequate as the reaction substrate and it is impossible to determine the activity of the enzyme by using uric acid.

		_										-		į.					
5			15	၉	7.5	10.2	0	0	3.5	S	우	0	0						
			14	2.75	6.9	9.3	0	0	3.3	4.5	6	0	0						
15		13	2.4	6.4	8.8	0	0	ဗ	4	8	0	0							
			12	2.1	5.8	7.9	0	0	2.6	3.5	7.4	0	0						
			Ŧ	1.87	5.3	7.1	0	0	2.3	3.2	6.8	0	0						
			10	1.56	4.7	6.3	0	0	1.9	2.9	9	0	0						
20			6	1.35	4.1	5	0	0	1.6	2.6	5.5	0	0						
25		(sər	ω	1.2	3.5	4.2	0	0	1.25	2.2	4.8	0	0						
		DO vali	7		2.8	3.4	0	0	-	1.9	4.3	0	0						
30		اues (∆	∇) sent	η sənlı	η sənı	∇) sente	p) sənا⊑	9	6.0	2.2	2.2	0	0	0.83	1.4	3.6	0	0	
		sured va	2	9.0	1.5	1.7	0	0	0.63	-:	က	0	0						
35	•	Table 1 Measured values (∆DO values)	4	9.0	0.8	=	0	0	0.5	0.75	2.2	0	0						
		Table	6	0.45	0.4	9.0	0	0	0.4	0.5	- 5:	0	0						
40			2	0.3	0.2	0.3	0	0	0.23	0.3	6.0	0	0						
			-	0.1	0.005	0.03	0	0	0.1	0.1	0.5	0	0						
45				0	0	0	0	0	0	0	0	0	0	0	1				
50			Amount of Enzyme (wt%)	(a)	(q)	(0)	(d) comparison	(e) comparison	(£)	(b)	(h)	(i) comparison	(i) comparison						
55			۱Å				1						1						

[0127] The weak reducing agent used in the present invention implies any reducing agent whose pseudo first-order reaction rate constant K_{obs} (as an index of reducing capacity) is smaller than 0.001 s⁻¹ (measured by the following method).

5 <Method for measurement of pseudo first-order reaction rate constant K_{obs}>

[0128] A solution is prepared which contains 300 mM of the reducing agent to be measured and 100 mM of phosphate buffer (pH = 5). To this solution is added a legal dye (violet No. 401) in an amount of 0.2 mM. The resulting solution is measured for the change of peak with time at the wavelength of 575 nm by using a UV meter (Model UV-160 made by Shimadzu Corporation). The rate constant for change with time is obtained, and it is defined as the pseudo first-order reaction rate constant $K_{\rm obs}$.

[0129] The above-mentioned weak reducing agent includes, for example, sodium thiosulfate, dl-cysteine, N-acetyl-L-cysteine, thiourea, dithioglycolic acid, L-ascorbic acid, sorbic acid, adipic acid, and salts thereof. They may be used alone or in combination with one another. The value of K_{obs} obtained by the above-mentioned method is as follows: 0.0081 for thioglycolic acid and 0.0068 for sodium sulfite (as strong reducing agents)

0.00075 for sodium thiosulfate, 0.00031 for N-acetyl-L-cysteine, and 0.00045 for thiourea (as weak reducing agents mentioned above)

[0130] The amount of the above-mentioned weak reducing agent in the composition is not specifically restricted; it may be adequately selected according to the kind of the weak reducing agent. It is usually equal to or less than 10%, preferably 0.01 to 10%, more preferably 0.1 to 7%. An excess amount may adversely affect dyeing power on account of excessively strong reducing action. The desired effect may not be produced if the amount is too small.

[0131] The composition of the present invention may be effective if containing cyclodextrin in addition to the above-mentioned essential ingredients. The cyclodextrin includes cyclodextrin and derivatives thereof. Cyclodextrin is a non-reducing maltoligosaccharide which has the structure in which 6 to 8 glucose molecules are connected to form a ring through the $(\alpha-1,4$ -glucoside bond. It takes the α -form, β -form, or γ -form depending on the number of glucose molecules connected together. The especially effective cyclodextrin derivative is one which is obtained by adding propylene oxide to said cyclodextrin. The number of moles to be added is not specifically restricted. It is usually 3 to 8 for one molecule. More than one kind of cyclodextrin may be used according to the compound to be included.

[0132] The amount of the above-mentioned cyclodextrin in the composition is not specifically restricted; it is usually 0.1 to 75%, preferably 0.5 to 60%. Cyclodextrin does not fully produce its effect if its amount is too small, and it does not produce its effect in proportion to its amount if its amount is too large.

[0133] The composition for dyeing keratinous fiber according to the present invention may be incorporated with additional components as listed below according to need so long as they do not prevent the effect of the present invention.

[0134] Acid, alkali as pH adjustor, surface active agent, ionic or nonionic natural or synthetic or semisynthetic polymeric compound, ester oil, vegetable oil, silicone derivative, fluorine derivative, amino acid, salts, alcohol (as solvent), dandruff remover, chelating agent, preservatives, UV absorber, fungicide, antioxidant, perfume, acid dye, and natural dye. These components are conventionally used ones, and they are added in an ordinary amount not harmful to the effect of the present invention.

[0135] The composition for dyeing keratinous fiber according to the present invention can be prepared in one-pack form by mixing the above-mentioned components in the usual way for dissolution, dispersion, or emulsification. A liquid solution type is desirable; but other types are also available such as paste (cream), aerosol, gel, liquid, and foam. The resulting preparations can be applied to hair, eyebrow, eyelash, and body hair to dye them as desired.

[0136] The invention will be described in more detail with reference to the following examples and comparative examples, which are not intended to restrict the scope of the invention.

[Examples 1 to 12 and Comparative Examples 1 to 6]

[0137] A stock solution for hair dye was prepared by uniformly mixing in the usual way with the components shown in Tables 2 and 3. The stock solution was placed in a glass pressure bottle, which was subsequently clinched under vacuum. The bottle was charged with LPG (2.0 kg) as a propellant in such an amount that the ratio of stock solution to gas is 95:5 (by mass). In this way there were obtained aerosol-type hair dyes (compositions for dyeing keratinous fiber) in Examples 1 to 12 and Comparative Examples 1 to 6. Each sample was examined for storage stability and dyeing property. The results are shown in Table 2 and 3.

(Test for storage stability)

[0138] Each sample was stored for six months at room temperature and for one month at 45°C. After storage, each

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sample was visually examined for aggregates, precipitates, and discoloration, and the results were rated according to the following criterion.

Rating criterion

[0139]

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- (in the contraction of the contr
- O: Aggregates, precipitates, and discoloration slightly discernible
- Δ : Aggregates, precipitates, and discoloration apparently discernible
- imes : Not usable due to excessive aggregates, precipitates, and discoloration

<Test for hair dyeing>

15 [0140] A bundle of goat white hair (about 10 g) was shampooed and dried. This hair was uniformly and rapidly coated with each sample (3 g) shown in Tables 2 and 3. After standing for about 20 minutes, the bundle of dyed hair was rinsed with running warm water and then shampooed and dried. The bundle of dried hair was measured for hair dyeing index (ΔΕ) by using a color difference meter (SE 2000 made by Nippon Denshoku Co., Ltd.) The hair dyeing index (ΔΕ) is obtained by measurement of L, a, b values of the dyed hair and calculating color difference (ΔΕ) from the undyed hair. This measurement was carried out immediately after preparation and after storage under the prescribed conditions mentioned above, so as to investigate the effect of storage. The larger the value of ΔΕ, the better the dyeing property.

Table 2

					נ	'abı	e_2						•	
5	<u> </u>		<u> </u>	Examples						<u> </u>				
			1	2	3	4	5	6	7	8	9	10	11	12
		Toluene-2,5- diamine sulfate	2.0	2.0			1.5	1.5	2.0	2.0	2.0	2.0	2.0	2.0
10		p-phenylenediamine			1.5	1.5	0.5	0.5	1.5	1.5				
10		Resorcin	0.5	0.5	0.75	0.75	0.25	0.25			1.5	1.5		
		Nitro-p- phenylenediamine	1.0	1.0			0.1	0.1						
15		p-nitro-o- phenylenediamine										•	1.5	1.5
		2,6- diaminopyridine			2.0	2.0	0.3	0.3	1.0	1.0	0.75	ð.75	0.5	0.5
	1	Lauric acid	0.3	0.3	0.3	0.3	0.3	0.3	0.3	0.3	0.3	0.3	0.3	0.3
20	Composition	Coconut oil fatty acid diethanolamine	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0
	1 #	Sorbitan laurate	0.5	0.5	0.5	0.5	0.5	0.5	0.5	0.5	0.5	0.5	0.5	0.5
	ğ	Ethanol	10	10	10	10	10	10	10	10	10	10	10	10
25	8	Hydroxyethylcell ulose	2.5	2.5	2.5	2.5	2.5	2.5	,2.5	2.5	2.5	2.5	2.5	2.5
25		Lactic acid	0.2	0.2	0.2	0.2	0.2	0.2	0.2	0.2	0.2	0.2	0.2	0.2
		β-cyclodextrin		1.0		1.0		1.2	L	1.0		1.5	L	1.0
		Laccase ⁽³⁾	7.01	7.01	7.01	7.01			3.52	3.5 ²			7.01	7.01
		Catechol oxidase(4)					7.01	7.01	3.52	3.5 ²	7.0 ¹	7.01		
30		N-acetyl-L- cysteine	1.0	1.0	1.0	1.0	1.0	1.0			1.0	1.0		
		Thiourea		Ĺ	L	<u> </u>	L	L	0.3	0.3	<u> </u>	<u> </u>	0.3	0.3
•		Monoethanolamine				Eno	ugh t	o adj	ust t	o pH	7.0			
35		Purified water							ance					
		Total (mass%)							0.0					
	۸ و	Initial	0	0	0	0	0	0	0	0	0	0	0	0
	Appearance, Storage stability	After storage for 6 months at r.t.	©- O	0	© -O	0	© -O	0	©- O	0	©- O	0	© -0	0
40	Appea Sto stab	After storage for 1 month at 45°C	©-O	0	©- O	0	©- O	0	© -O	0	©- O	0	0 -0	0
	loa, y	Initial (initial color)	(x)	Q F	(1b)	(1b)	(rb)	(rb)	© (ly)	(1y)	(1b)	(1b)	(dy)	(dy)
45	Discoloration Storage Stability	After storage for 6 months at r.t.	©-O	0	0 -0	0	©- O	0	© -0	0	© -O	0	©- O	0
	Disco S St	After storage for 1 month at 45°C	©-0	0	©- O	0	© -O	0	©- O	0	©- O	0	© -0	0
50	·	Dyed color	1b	1b	nb	nb	ďЪ	đb	bk	bk	bk	bk	У	У
	ing srty E	Initial	31.0	31.5	30.2	30.9	32.1	32.0	35.5	35.0	33.1	33.5		28.5
55	Dyeing Property, ΔE	After storage for 1 month at 45°C	30.1	30.8	29.7	29.8	30.2	30.4	32.1	32.3	30.0	30.3	27.4	27.4
33	·					نـــــــــــــــــــــــــــــــــــــ								L

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	Note to Table 2.
	7.0\(^1\): equivalent to Δ DO = 1.0
5	3.5^2 : equivalent to $\Delta DO = 0.5$
	Laccase ⁽³⁾ : with an active value equivalent to $\Delta DO = 15.3$
10	Catechol amine $^{(4)}$: with an active value equivalent to ΔDO =
	15.5
	Color: r = red, lb = light brown, ly = light yellow, dy =
	dark yellow, nb = navy blue, db = dark brown, bk = black, y
15	= yellow.
	r.t. = room temperature
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Table 3

					Comparativ	e Examples	· ·		
			1	2	3	4	5	6	
	Composition	Toluene-2,5-diamine sulfate	2.0	2.0	2.0	2.0	2.0	2.0	
		p-phenylenediamine							
		Resorcin	0.5	0.5	0.5	0.5	0.5	0.5	
		Nitro-p- phenylenediamine			•				
		p-nitro-o- phenylenediamine							
		2,6-diaminopyridine	1.0	1.0	1.0	1.0	1.0	1.0	
		Lauric acid	0.3	0.3	0.3	0.3	0.3	0.3	
		Coconut oil fatty acid diethanolamine	1.0	1.0	1.0	1.0	1.0	1.0	
		Sorbitan laurate	0.5	0.5	0.5	0.5	0.5	0.5	
		Ethanol	10	10	10	10	10	10	
		Hydroxyethylcellulose	2.5	2.5	2.5	2.5	2.5	2.5	
		Lactic acid	0.2	0.2	0.2	0.2	0.2	0.2	
5		β-cyclodextrin		1.0			1.0	1.0	
		Laccase (3)				7.0 ¹		7.0 ¹	
		Catechol oxidase			***				
		N-acetyl-L-cysteine			1.0		1.0		
		Thiourea							
		Monoethanolamine		E	nough to ad	ljust to pH 7	'.0		
		Purified water	Balance						
		Total (mass%)			10	0.0			
	Appearance,	Initial	0	0	0	0	0	0	
	Storage stability	After storage for 6 months at r.t.	0	0	0	×	0	0	
		After storage for 1 month at 45°C	0	0	0	×	0	0	
	Discoloration,	Initial (initial color)	⊚ (ly)	⊚ (ly)	⊚ (ly)	⊚ (ly)	⊚ (ly)	⊚ (ly)	
	Storage Stability	After storage for 6 months at r.t.	0	0	0	×	0	Δ	
		After storage for 1 month at 45°C	0	0	0	×	0	×	
	Dyeing Property,	Dyed color	not dyed	not dyed	not dyed	lb	not dyed	lb	
	ΔΕ	Initial	5.4	6.6	5.0	29.9	4.5	29.5	
		After storage for 1 month at 45°C	3.2	4.2	3.8	8.4	3.2	21.8	

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Note to Table 3.
7.0¹ : equivalent to $\Delta DO = 1.0$ Laccase⁽³⁾ : with an active value equivalent to $\Delta DO = 15.3$ Color: ly : light yellow, lb : light brown r.t. = room temperature

[0141] It is noted from Tables 2 and 3 that the composition for dyeing keratinous fiber (reactive hair dye) according to the present invention exhibits very good storage stability and dyeing property. It exhibits better stability and dyeing property if it is additionally incorporated with β-cyclodextrin. By contrast, the samples in Comparative Examples 1, 2, 3, and 5, which do not contain the oxidase specified in the present invention, are stable but incapable of dyeing because of the lack of oxidizing ability. The samples in Comparative Examples 4 and 6, which do not contain the weak reducing agent specified in the present invention, are capable of dyeing immediately after preparation but deteriorate with aggregates, precipitates, and discoloration after storage for six months at room temperature or for one month at 45°C. They are also poor in dyeing property.

[0142] Specific examples of the composition are shown below, although they are not intended to restrict the scope of the present invention. The laccase used in the following examples is one which has an active value equivalent to $\Delta DO = 15.3$.

[Example 13] One-pack hair dye (in the form of foam)

[0143]

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Components	Amount (mass%)
(Stock solution)	
p-phenylenediamine	1.0
2,5-diaminotoluene sulfate	2.0
m-phenylenediamine	0.5
p-aminophenol	0.5
2-(2'-hydroxyethylamino)-5-aminotoluene sulfate	0.15
Oleic acid	0.2
Oleyl alcohol	0.2
β-cyclodextrin	1.0
Laccase	7.0(1)
Sodium polyacrylate (cross-linked type)	1.0
Cationized hydroxyethyl cellulose	0.5
N-acetyl-L-cysteine	1.0
Sorbitan monolaurate	1.0
Ethanol	10.0
Glycolic acid	0.2
Perfume	0.1
Purified water	balance
(Adjusted to pH 6.5 with monoethanolamine)	
Total	100

Laccase $7.0^{(1)}$; equivalent to $\Delta DO = 1.0$

[0144] A stock solution was prepared in the usual way according to the above-mentioned formulation. The stock solution was placed in an aerosol can, which was subsequently clinched under vacuum. The aerosol can was filled with LPG (2.0 kg) as a propellant such that the ratio of stock solution to gas is 95:5 (by mass). Thus there was obtained a hair dye of aerosol type. The LPG as a compressed gas may be replaced by any one or more of nitrogen, carbon dioxide gas, dinitrogen monoxide gas, flon 11, flon 12, and flon 114. The aerosol type may be direct spray type or piston type in an aluminum can or tinplate can, or the aerosol can may be double-walled can such as back-in type and EXXEL type. After storage for six months at room temperature or for one month at 45°C, the above-mentioned hair dye was applied in an adequate amount to 1.0 g of white hair, and after standing for 20 minutes, the hair was washed with water and shampooed. The treated hair was dried by a drier. It was found that the hair was dyed in blue black. This color was the same as that obtained when the hair dye was used immediately after production.

Amount (mass%)

Components

[Example 14] One-pack hair dye (in the form of foam)

(Stock solution)

[0145]

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5,6-dihydroxyindoline hydrobromide 1.0 5,6-dihydroxyindole hydrochloride 1.0 N-ethyl-5,6-dihydroxyindole hydrochloride 0.05 Linoleic acid 0.2 Oleyl alcohol 0.2 **B**-cyclodextrin 1.0 Laccase 7.0(1) Hydroxyethyl cellulose 0.5 Coconut oil fatty acid sodium acyl-glutamate 1.0 N-acetyl-L-cysteine 0.5 Thiourea 0.3 Sodium polyacrylate (cross-linked type) 0.2 Ethanol 10.0 Lactic acid 0.2 Perfume 0.1 Purified water balance (Adjusted to pH 6.8 with monoethanolamine) Total 100

Laccase $7.0^{(1)}$: equivalent to $\Delta DO = 1.0$

[0146] A stock solution was prepared in the usual way according to the above-mentioned formulation. The stock solution was placed in an aerosol can, which was subsequently clinched under vacuum. The aerosol can was filled with LPG (2.0 kg) as a propellant such that the ratio of stock solution to gas is 95:5 (by mass). Thus there was obtained a hair dye of aerosol type. After storage for six months at room temperature or for one month at 45°C, the abovementioned hair dye was applied (in an adequate amount) to 1.0 g of white hair, and after standing for 20 minutes, the hair was washed with water and shampooed. The treated hair was dried by a drier. It was found that the hair was dyed in black. This color was the same as that obtained when the hair dye was used immediately after production.

[Example 15] One-pack hair dye (in the form of creamy foam)

[0147]

Components	Amount (mass%)
(Stock solution)	1
2,5-diaminotoluene sulfate	2.0
m-phenylenediamine	0.5
m-aminophenol	0.5
Resorcin	0.1
o-aminocresol	0.05
Oleic acid	0.2
β-cyclodextrin	1.2
Laccase	10.5(2)
Xanthan gum	0.05
Sodium sulfite	0.05
Thiourea	0.05
N-acetyl-L-cysteine	0.5

(continued)

Components	Amount (mass%)
(Stock solution)	
Stearyl trimethylammonium chloride	0.2
Cetostearyl alcohol	0.6
POE (20) hardened castor oil triisostearate	0.2
Sorbitan monostearate	0.1
Methylparaben	0.3
Glycolic acid	0.2
Propylene glycol	5.0
Perfume	0.1
Purified water	balance
(Adjusted to pH 7.0 with monoethanolamine)	
Total	100

Laccase 10.5(2): equivalent to ΔDO = 1.5

[0148] A stock solution was prepared in the usual way according to the above-mentioned formulation. The stock solution was placed in an aerosol can, which was subsequently clinched under vacuum. The aerosol can was filled with LPG (4.0 kg) as a propellant such that the ratio of stock solution to gas is 95:5 (by mass). Thus there was obtained a creamy hair dye of aerosol type. After storage for six months at room temperature or for one month at 45°C, the above-mentioned hair dye was applied in an adequate amount to 1.0 g of white hair, and after standing for 20 minutes, the hair was washed with water and shampooed. The treated hair was dried by a drier. It was found that the hair was dyed in dark brown. This color was the same as that obtained when the hair dye was used immediately after production.

[Example 16] One-pack hair dye (in the form of creamy foam)

[0149]

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Components	Amount (mass%)
(Stock solution)	
5,6-dihydroxyindoline hydrobromide	1.0
5,6-dihydroxyindole hydrochloride	1.0
N-methyl-5,6-dihydroxyindole hydrobromide	0.05
Linoleic acid	0.2
β-cyclodextrin	1.0
Laccase	10.5 ⁽²⁾
Sodium sulfite	0.09
N-acetyl-L-cysteine	0.3
Cetostearyl trimethylammonium chloride	0.2
Cetostearyl alcohol	0.6
POE (20) hardened castor oil monoisostearate	0.2
Sorbitan monostearate	0.1
Methylparaben	0.3
Lactic acid	0.2
Diethylene glycol monoethyl ether	5.0
1,3-butylene glycol	3.0
Perfume	0.1
Purified water	balance
(Adjusted to pH 7.0 with monoethanolamine)	
Total	100

Laccase 10.5⁽²⁾: equivalent to ΔDO = 1.5

[0150] A stock solution was prepared in the usual way according to the above-mentioned formulation. The stock solution was placed in an aerosol can, which was subsequently clinched under vacuum. The aerosol can was filled with LPG (4.0 kg) as a propellant such that the ratio of stock solution to gas is 95:5 (by mass). Thus there was obtained a hair dye of aerosol type. After storage for six months at room temperature or for one month at 45°C, the above-mentioned hair dye was applied in an adequate amount to 1.0 g of white hair, and after standing for 20 minutes, the hair was washed with water and shampooed. The treated hair was dried by a drier. It was found that the hair was dyed in black. This color was the same as that obtained when the hair dye was used immediately after production.

[Example 17] One-pack hair dye (cream type)

[0151]

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Components	Amount (mass%)
5,6-dihydroxyindoline hydrobromide	1.0
5,6-dihydroxyindole hydrochloride	₫.0
N-methyl-5,6-dihydroxyindole hydrobromide	0.05
N-methyl-5,6-dihydroxyindole hydrochloride	0.05
2,5-diaminotoluene sulfate	0.01
β-cyclodextrin	1.0
Laccase	14.0 ⁽³⁾
N-acetyl-L-cysteine	0.5
Thiourea	. 0.3
Alkyl trimethylammonium chloride	0.5
Coconut oil fatty acid acyl-L-arginine ethyl-D,L-pyrrolidone carboxylate	0.5
Cetostearyl alcohol	2:0
Oleyi alcohol	1.0
POE (40) hardened castor oil	0.75
POE (20) stearyl ether	0.75
Sorbitan sesquistearate	1.0
Methylparaben	0.3
Propylene glycol	5.0
Glycolic acid	0.2
Perfume	0.1
Purified water	balance
(Adjusted to pH 7.5 with monoethanolamine)	
Total	100

Laccase $14.0^{(3)}$: equivalent to $\Delta DO = 2.0$

[0152] The above-mentioned components were uniformly mixed in the usual way to give a hair dye of cream type. After storage for six months at room temperature or for one month at 45°C, the above-mentioned hair dye was applied (in an adequate amount) to 1.0 g of white hair, and after standing for 20 minutes, the hair was washed with water and shampooed. The treated hair was dried by a drier. It was found that the hair was dyed in dark brown. This color was the same as that obtained when the hair dye was used immediately after production.

[Example 18] One-pack hair dye (cream type)

[0153]

 Components
 Amount (mass%)

 2,5-diaminotoluene sulfate
 2.0

 2,6-diaminopyridine
 0.05

 N,N-bis(β-hydroxyl)-p-phenylenediamine
 0.1

 2-amino-5-orthophenol
 0.5

(continued)

Components	Amount (mass%)
2-(2'-hydroxyethylamino)-5-aminotoluene	0.15
Linoleic acid	0.2
β-cyclodextrin	1.0
Laccase	14.0 ⁽³⁾
N-acetyl-L-cysteine	1.0
Thiourea	0.5
Stearyl trimethylammonium chloride	0.5
Behenyl trimethylammonium chloride	0.5
Cetostearyl alcohol	2.0
Oleyi alcohol	1.0
POE (40) glycerin triisostearate	0.75
POE (20) lauryl ether	0.75
Sorbitan monostearate	1.0
Methylparaben	0.3
Propylene glycol	5.0
Lactic acid	0.2
Perfume	0.1
Purified water	balance
(Adjusted to pH 7.0 with monoethanolamine)	2
Total	100.0

Laccase $14.0^{(3)}$: equivalent to $\Delta DO = 2.0$

[0154] The above-mentioned components were uniformly mixed in the usual way to give a hair dye of cream type.

After storage for six months at room temperature or for one month at 45°C, the above-mentioned hair dye was applied in an adequate amount to 1.0 g of white hair, and after standing for 20 minutes, the hair was washed with water and shampooed. The treated hair was dried by a drier. It was found that the hair was dyed in light brown. This color was the same as that obtained when the hair dye was used immediately after production.

[Example 19] One-pack hair dye (treatment type)

[0155]

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Components	Amount (mass%)
2.5-diaminotoluene sulfate	5.0
2-amino-4-nitrophenol	3.0
5-amino-o-cresol	1.0
p-aminophenol	1.0
Oleic acid	0.5
Linoleic acid	0.5
β-cyclodextrin	2.0
Laccase	14.0(3)
N-acetyl-L-cysteine	1.0
Thiourea	0.5
Stearyl trimethylammonium chloride	0.5
Cetyl trimethylammonium chloride	0.5
Cetostearyl alcohol	4.0
Oleyl alcohol	1.0
Ethyl oleate	0.5
Isopropyl palmitate	0.5

(continued)

Components	Amount (mass%)
Liquid paraffin	1.0
Beeswax	0.5
POE (40) hardened castor oil triisostearate	0.25
POE (20) hardened castor oil triisostearate	0.25
POE (30) stearyl ether	0.75
Sorbitan monostearate	1.0
Glycerin monostearate	0.5
Methylparaben . •	0.3
Propylene glycol	5.0
Glycerin	3.0
Glycolic acid	0.2
Perfume	0.1
Purified water	balance
(Adjusted to pH 8.0 with monoethanolamine)	
Total	100.0

Laccase $14.0^{(3)}$: equivalent to $\Delta DO = 2.0$

[0156] The above-mentioned components were uniformly mixed in the usual way to give a hair dye of treatment type. After storage for six months at room temperature or for one month at 45°C, the above-mentioned hair dye was applied in an adequate amount to 1.0 g of white hair, and after standing for 20 minutes, the hair was washed with water and shampooed. The treated hair was dried by a drier. It was found that the hair was dyed in red brown. This color was the same as that obtained when the hair dye was used immediately after production. In addition, this hair dye gave good hand to the treated hair and produced good treatment effect.

[Example 20] One-pack hair dye (treatment type)

[0157]

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Components	Amount (mass%)
5,6-dihydroxyindoline hydrobromide	1.0
5,6-dihydroxyindole hydrochloride	1.0
N-methyl-5,6-dihydroxyindoline hydrobromide	0.5
N-methyl-5,6-dihydroxyindole hydrochloride	0.5
5-aminoindole hydrochloride	0.25
2,3-dimethyl-5,6-dihydroxyindoline hydrobromide	0.25
Oleic acid	0.5
Linoleic acid	0.5
β-cyclodextrin	2.0
Laccase	14.0 ⁽³⁾
N-acetyl-L-cysteine	1.0
Thiourea	0.5
Stearyl trimethylammonium chloride	0.5
Cetyl trimethylammonium chloride	0.5
Cetostearyl alcohol	4.0
Oleyl alcohol	1.0
Ethyl oleate	0.5
Isopropyl palmitate	0.5
Liquid paraffin	1.0

Laccase $14.0^{(3)}$: equivalent to $\Delta DO = 2.0$

(continued)

Components	Amount (mass%)
Beeswax	0.5
POE (40) hardened castor oil triisostearate	0.25
POE (20) hardened castor oil triisostearate	0.25
POE (30) stearyl ether	0.75
Sorbitan monostearate	1.0
Glycerin monostearate	0.5
Methylparaben	0.3
Propylene glycol	5.0
1,3-butylene glycol	3.0
Lactic acid	0.2
Perfume	0.1
Purified water	balance
(Adjusted to pH 7.5 with monoethanolamine)	
Total	100.0

[0158] The above-mentioned components were uniformly mixed in the usual way to give a hair dye of treatment type. After storage for six months at room temperature or for one month at 45°C, the above-mentioned hair dye was applied in an adequate amount to 1.0 g of white hair, and after standing for 20 minutes, the hair was washed with water and shampooed. The treated hair was dried by a drier. It was found that the hair was dyed in dark gray black. This color was the same as that obtained when the hair dye was used immediately after production. In addition, this hair dye gave good hand to the treated hair and produced good treatment effect.

[Example 21] One-pack hair dye (gel type)

³⁰ [0159]

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Components	Amount (mass%)
p-phenylenediamine	1.0
2,5-diaminotoluene sulfate	2.0
m-phenylenediamine	0.5
p-aminophenol	0.5
2-(2'-hydroxyethylamino)-5-aminotoluene sulfate	0.15
Oleic acid	0.2
β-cyclodextrin	1.0
Laccase	5.5 ⁽⁴⁾
Xanthan gum	0.5
N-acetyl-L-cysteine	0.1
Hydroxyethyl cellulose	1.0
POE (40) lauryl ether	1.0
POE (30) hardened castor oil	1.0
Ethanol	10.0
Glycolic acid	0.2
Perfume	0.1
Purified water	balance
(Adjusted to pH 6.5 with monoethanolamine)	
Total	100.0

Laccase $5.5^{(4)}$: equivalent to $\triangle DO = 0.8$

[0160] The above-mentioned components were uniformly mixed in the usual way to give a hair dye of gel type. After

storage for six months at room temperature or for one month at 45°C, the above-mentioned hair dye was applied (in an adequate amount) to 1.0 g of white hair, and after standing for 20 minutes, the hair was washed with water and shampooed. The treated hair was dried by a drier. It was found that the hair was dyed in blue black. This color was the same as that obtained when the hair dye was used immediately after production.

[Example 22] One-pack hair dye (gel type)

[0161]

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}	20	
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Components Amount (mass%)		
5,6-dihydroxyindoline hydrobromide	1.0	
5,6-dihydroxyindole hydrochloride	1.0	
N-ethyl-5,6-dihydroxyindole hydrobromide	0.05	
Oleic acid	0.2	
β-cyclodextrin	1.0	
Laccase	5.5 ⁽⁴⁾	
Xanthan gum	0.5	
N-acetyl-L-cysteine	1.0	
Hydroxyethyl cellulose	1.0	
POE (40) lauryl ether	- 1.0	
POE (30) hardened castor oil	1.0	
Ethanol	10.0	
Glycolic acid	0.2	
Perfume .	0.1	
Purified water	balance	
(Adjusted to pH 6.8 with monoethanolamine)		
Total	100.0	

Laccase $5.5^{(4)}$: equivalent to $\Delta DO = 0.8$

[0162] The above-mentioned components were uniformly mixed in the usual way to give a hair dye of gel type. After storage for six months at room temperature or for one month at 45°C, the above-mentioned hair dye was applied (in an adequate amount) to 1.0 g of white hair, and after standing for 20 minutes, the hair was washed with water and shampooed. The treated hair was dried by a drier. It was found that the hair was dyed in black. This color was the same as that obtained when the hair dye was used immediately after production.

[Example 23] One-pack eyelash dye (gel type)

[0163]

2,5-diaminotoluene sulfate 2.0)
Nitro-p-phenylenediamine 1.5	5
p-aminophenol 0.5	5
m-aminophenol 0.3	3
Sodium oleate 0.2	2
50 β-cyclodextrin 1.0)
Laccase .3.50	5)
N-acetyl-L-cysteine 0.5	5
Thiourea 0.3	3
Hydroxyethyl cellulose 0.1)
C ₁₀ polycarbamyl polyglycol ester 0.1]
POE (40) lauryl ether)

Laccase $3.5^{(5)}$: equivalent to $\Delta DO = 0.5$

(continued)

Components	Amount (mass%)
Ethanol	10.0
Glycolic acid	. 0.2
Perfume	0.1
Purified water	balance
(Adjusted to pH 8.0 with monoethanolamine)	
Total	100.0

[0164] The above-mentioned components were uniformly mixed in the usual way to give an eyelash dye of gel type. After storage for six months at room temperature or for one month at 45°C, the above-mentioned eyelash dye was evaluated by ten female panelists. All the panelists had their eyelash dyed apparently in red. It was found that the eyelash dye kept its dyeing power even after storage. Thus this eyelash dye was rated as excellent.

[Example 24] One-pack eyebrow dye (soft cream type)

[0165]

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	Components	Amount (mass%)
	2.5-diaminotoluene sulfate	2.0
	m-phenylenediamine	2.5
25	Oleic acid	0.5
	Linoleic acid	0.5
	β-cyclodextrin	2.0
	Laccase	10.5 ⁽⁶⁾
	N-acetyl-L-cysteine	0.5
30	Thiourea	0.3
	Cetostearyl alcohol	2.5
	POE (3) lauryl ether sulfate	0.8
	Oleyl alcohol	1.0
35	Alkyl acrylate alkyl methacrylate polyoxyethylene (20) stearyl ether copolymer	1.0
55	Isopropyl myristate	0.5
	POE (40) glycerin isostearate	0.25
	POE (20) glycerin isostearate	0.25
	Sorbitan sesquioelate	1.0
40	Methylparaben	. 0.3
	Propylene glycol	5.0
	Glycolic acid	0.2
	Perfume	0.1
	Purified water	balance
45	(Adjusted to pH 7.5 with monoethanolamine)	
	Total	100.0

Laccase $10.5^{(6)}$: equivalent to $\Delta DO = 1.5$

[0166] The above-mentioned components were uniformly mixed in the usual way to give an eyebrow dye of cream type. After storage for six months at room temperature or for one month at 45°C, the above-mentioned eyebrow dye was evaluated by ten female panelists. All the panelists had their eyebrow dyed apparently in blue. It was found that the eyebrow dye kept its dyeing power even after storage. Thus this eyebrow dye was rated as excellent.

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[Example 25] One-pack hair dye (cream type)

[0167]

5	Components Amount (mass%)	Components Amount (mass%)		
	2,5-diaminotoluene sulfate	1.0		
	HC red No. 1	1.0		
	HC red No. 16	0.5		
10	4-amino-m-cresol	0.2		
	1,3-bis-(2,4-diaminophenoxy)-propane .	0.5		
	4-chlororesorcinol · •	0.15		
	Linoleic acid	0.2		
	β-cyclodextrin	1.0		
15	Laccase	14.0(7)		
	N-acetyl-L-cysteine	0.8		
	Stearyl trimethylammonium chloride	0.2		
	Cetostearyl alcohol	1.5		
20	POE (40) glycerin triisostearate	0.75		
	POE (20) lauryl ether	0.5		
	POE (30) oleyl ether	0.5		
	Sorbitan monostearate	1.0		
	Methylparaben .	0.3		
25	Propylene glycol	5.0		
	Lactic acid	0.2		
	Perfume	0.1		
	Purified water	balance		
30	(Adjusted to pH 7.0 with monoethanolamine)			
	Total	100.0		

Laccase (7): equivalent to $\Delta DO = 2.0$

[0168] The above-mentioned components were uniformly mixed in the usual way to give a hair dye of cream type. After storage for six months at room temperature or for one month at 45°C, the above-mentioned hair dye was applied (in an adequate amount) to 1.0 g of white hair, and after standing for 20 minutes, the hair was washed with water and shampooed. The treated hair was dried by a drier. It was found that the hair was dyed in pretty dark red. This color was the same as that obtained when the hair dye was used immediately after production.

[Example 26] One-pack hair dye (cream type)

[0169] .

45	Components	Amount (mass%)
	2,5-diaminotoluene sulfate	1.0
	HC blue No. 2	1.0
	HC blue No. 7	0.5
	Hydroxyethyl-2-nitro-p-toluidine	0.25
50	4-amino-2-hydroxytoluene	0.15
	1,5-naphthalenediol	0.05
	2-amino-3-hydroxypyridine	0.05
•	β-cyclodextrin	1.0
55	Laccase	14.0(7)
	N-acetyl-L-cysteine	0.5
	Thiourea	0.3

(continued)

Components		Amount (mass%)
Stearyl trimethylammonium chloride		0.5
Coconut oil fatty acid acyl-L-arginine ethyl-D,L-pym	olidone carboxylate	0.3
Cetostearyl alcohol		2.5
Oleyl alcohol		0.5
POE (30) hardened castor oil		0.75
POE (30) cetyl ether		0.75
Sorbitan monostearate	•	1.0
Methylparaben	. 👂	0.3
Propylene glycol	•	2.0
1,3-butylene glycol		2,0
Glycolic acid		0.2
Perfume		δ.1
Purified water		balance
(Adjusted to pH 7.5 with monoethanolamine)		
Total		100.0

Laccase $14.0^{(7)}$: equivalent to $\Delta DO = 2.0$

[0170] The above-mentioned components were uniformly mixed in the usual way to give a hair dye of cream type. After storage for six months at room temperature or for one month at 45°C; the above-mentioned hair dye was applied (in an adequate amount) to 1.0 g of white hair, and after standing for 20 minutes, the hair was washed with water and shampooed. The treated hair was dried by a drier. It was found that the hair was dyed in deep blue. This color was the same as that obtained when the hair dye was used immediately after production.

[Example 27] One-pack hair dye (in the form of creamy foam)

[0171]

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Components	Amount (mass%)
(Stock solution)	
2,5-diaminotoluene sulfate	1.0
HC yellow No. 6	0.7
HC yellow No. 2	0.3
HC orange No. 1	0.7
2-amino-6-chloro-4-nitrophenol	- 0.3
2-amino-6-chloro-o-cresol	0.3
2,6-dihydroxyethylaminotoluene	0.1
Hydroxyethyl-2-nitro-p-toluidine	0.05
Linoleic acid	0.2
B-cyclodextrin	1.0
Laccase	10.5 ⁽⁸⁾
Sodium sulfite	0.09
N-acetyl-L-cysteine	0.3
Cetostearyl trimethylammonium chloride	0.2
Cetostearyl alcohol	0.6
POE (20) glycerin triisostearate	0.2
Sorbitan monostearate	0.1
POE (20) Oleyl ether	0.2
Methylparaben	0.3

Laccase $10.5^{(8)}$: equivalent to $\Delta DO = 1.5$

(continued)

Components	Amount (mass%)
(Stock solution)	
Lactic acid .	0.2
1,3-butylene glycol	5.0
Perfume	0.1
Purified water	balance
(Adjusted to pH 7.0 with monoethanolamine)	
Total	100.0

[0172] A stock solution was prepared in the usual way according to the above-mentioned formulation. The stock solution was placed in an aerosol can, which was subsequently clinched under vacuum. The aerosol can was filled with LPG (4.0 kg) as a propellant such that the ratio of stock solution to gas is 95:5 (by mass). Thus there was obtained a hair dye of aerosol type. After storage for six months at room temperature or for one month at 45°C, the above-mentioned hair dye was applied (in an adequate amount) to 1.0 g of white hair, and after standing for 20 minutes, the hair was washed with water and shampooed. The treated hair was dried by a drier. It was found that the hair was dyed in light golden color. This color was the same as that obtained when the hair dye was used immediately after production. In addition, when applied to European blond hair, this hair dye enhanced the blond color.

[Example 28] One-pack hair dye (in the form of foam)

[0173]

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Components	Amount (mass%)	
(Stock solution)		
2,5-diaminotoluene sulfate	2.0	
4-hydroxypropylamino-3-nitrophenol	0.7	
N,N'-bis-(2-hydroxyethyl)-2-nitro-para-henylenediamine	0.5	
2,7-naphthalenediol	0.3	
2-methylresorcinol	0.3	
Oleic acid	0.2	
Oleyl alcohol	0.2	
β-cyclodextrin	1.0	
Laccase	7.0 ⁽⁹⁾	
Cationized hydroxyethyl cellulose	0.5	
N-acetyl-L-cysteine	1.0	
Sorbitan sesquioleate	1.5	
Ethanol	10.0	
Glycolic acid	0.2	
Perfume	0.1	
Purified water	balance	
(Adjusted to pH 6.5 with monoethanolamine)		
Total	100.0	

Laccase $7.0^{(9)}$: equivalent to $\Delta DO = 1.0$

[0174] A stock solution was prepared in the usual way according to the above-mentioned formulation. The stock solution was placed in an aerosol can, which was subsequently clinched under vacuum. The aerosol can was filled with LPG (2.0 kg) as a propellant such that the ratio of stock solution to gas is 95:5 (by mass). Thus there was obtained a hair dye of aerosol type. The LPG as a compressed gas may be replaced by any one or more of nitrogen, carbon dioxide gas, dinitrogen monoxide gas, flon 11, flon 12, and flon 114. The aerosol type may be direct spray type or piston type in an aluminum can or tinplate can, or the aerosol can may be double-walled can such as back-in type and EXXEL

type. After storage for six months at room temperature or for one month at 45°C, the above-mentioned hair dye was applied in an adequate amount to 1.0 g of white hair, and after standing for 20 minutes, the hair was washed with water and shampooed. The treated hair was dried by a drier. It was found that the hair was dyed in blackish indigo blue. This color was the same as that obtained when the hair dye was used immediately after production.

[Example 29] One-pack hair dye (in the form of foam)

[0175]

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	Components	Amount (mass%)
	(Stock solution)	
	2,5-diaminotoluene sulfate	2.0
15	4-amino-2-nitiodiphenylamine-2'-carboxylic acid	1.0
	Tetrahydro-6-nitroquinoxaline	1.0
	2-aminomethyl-p-aminophenol hydrochloride	0.5
	2-amino-6-chloro-4-nitrophenol	0.3
	2,4-diaminophenoxyethanol hydrochloride	0.3
	Linoleic acid	0.2
•	Oleyl alcohol	0.2
	β-cyclodextrin	1.0
L -	Laccase	7.0 ⁽⁹⁾
25	Hydroxyethyl cellulose	0.5
	Coconut oil fatty acid sodium acyl-glutamate	1.0
	N-acetyl-L-cysteine	0.5
	Thiourea	0.3
	Sodium polyacrylate (cross-linked type)	2.5
	Ethanol	10.0
0	Propylene glycol	5.0
	Lactic acid	0.2
	Perfume	0.1
	Purified water	balance
25	(Adjusted to pH 6.8 with monoethanolamine)	
35	Total	100.0

Laccase $7.0^{(9)}$: equivalent to $\Delta DO = 1.0$

40 [0176] A stock solution was prepared in the usual way according to the above-mentioned formulation. The stock solution was placed in an aerosol can, which was subsequently clinched under vacuum. The aerosol can was filled with LPG (2.0 kg) as a propellant such that the ratio of stock solution to gas is 95.5 (by mass). Thus there was obtained a hair dye of aerosol type. After storage for six months at room temperature or for one month at 45°C, the above-mentioned hair dye was applied in an adequate amount to 1.0 g of white hair, and after standing for 20 minutes, the hair was washed with water and shampooed. The treated hair was dried by a drier. It was found that the hair was dyed in dark brown. This color was the same as that obtained when the hair dye was used immediately after production.

[Example 30] One-pack hair dye (in the form of creamy foam)

₅₀ [0177]

Components	Amount (mass%)	
(Stock solution)		
2,5-diaminetoluene sulfate	2.0	
2-amino-3-hydroxypyridine	1.0	
2,6-dihydroxyethylaminotoluene	0.5	

(continued)

Components	Amount (mass%)	
(Stock solution)		
4-amino-3-nitrophenol	0.3	
2,6-dihydroxy-3,4-dimethylpyridine	0.3	
Oleic acid	0.2	
β-cyclodextrin	1.2	
Laccase	10.5 ⁽¹⁰⁾	
Xanthan gum .	0.05	
Sodium sulfite	0.05	
Thiourea	0.05	
N-acetyl-L-cysteine	0.2	
Stearyl trimethylammonium chloride	0.2	
Cetostearyl alcohol	1.0	
N-lauroyl-N-methyl-β-alanine triethanolamine	0.5	
POE (20) glycerin triisostearate	· 0.2	
Sorbitan monooleate	0.2	
Methylparaben	0.3	
Glycolic acid	0.2	
Propylene glycol	15.0	
Perfume	2 0.1	
Purified water	balance	
(Adjusted to pH 7.0 with monoethanolamine)		
Total	100.0	

Laccase $10.5^{(10)}$: equivalent to $\Delta DO = 1.5$

[0178] A stock solution was prepared in the usual way according to the above-mentioned formulation. The stock solution was placed in an aerosol can, which was subsequently clinched under vacuum. The aerosol can was filled with LPG (4.0 kg) as a propellant such that the ratio of stock solution to gas is 95.5 (by mass). Thus there was obtained a creamy hair dye of aerosol type. After storage for six months at room temperature or for one month at 45°C, the above-mentioned hair dye was applied (in an adequate amount) to 1.0 g of white hair, and after standing for 20 minutes, the hair was washed with water and shampooed. The treated hair was dried by a drier. It was found that the hair was dyed in dark brown. This color was the same as that obtained when the hair dye was used immediately after production. [0179] The composition for dyeing keratinous fiber according to the present invention is incorporated with a weak reducing agent as mentioned above, so that it is available in one-pack form (in place of a mixed type which needs the mixing of two packs before use) for consumer's convenience. Even though it is of one-pack type, it remains stable without aggregates, precipitates, and discoloration during storage at high temperatures. Its stability is enhanced by incorporation with cyclodextrin. It produces its effect through reaction between oxidase and oxidative color-developing substance regardless of its form (aerosol, cream, gel, and liquid).

Claims

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- A composition for dyeing keratinous fiber which comprises incorporated therein an oxidative color-developing substance, an enzyme which reacts with oxygen as a substrate but does not evolve hydrogen peroxide, and a weak reducing agent.
- 2. The composition for dyeing keratinous fiber as defined in Claim 1, which further comprises cyclodextrin.

INTERNATIONAL SEARCH REPORT

International application No.
PCT/JP00/08525

A. CLASSII	A. CLASSIFICATION OF SUBJECT MATTER				
Int.C	A. CLASSIFICATION OF SUBJECT WATTER Int.Cl ⁷ A61K7/13, C09B53/00, C09B55/00, C09B57/00				
ading to 1	According to International Patent Classification (IPC) or to both national classification and IPC				
a region	CEARCHED				
	B. FIELDS SEARCHED Minimum documentation searched (classification system followed by classification symbols) Int.Cl ⁷ A61K7/13, C09B53/00, C09B55/00, C09B57/00				
5	on searched other than minimum documentation to the ex-	tent that such documents are included i	n the fields searched		
			a		
Electronic data base consulted during the international search (name of data base and, where practicable, search terms used) CAPLUS (STN), JOIS (JICST)					
C. DOCUM	MENTS CONSIDERED TO BE RELEVANT				
Category*	Citation of document, with indication, where appro-	opriate, of the relevant passages	Relevant to claim No.		
PX PA	WO, 00/57848, Al (L'oreal), 05 October, 2000 (05.10.00), Claims & FR, 2791256, Al	i	2		
PX PA	EP, 1043012, A2 (L'oreal), 11 October, 2000 (11.10.00), Claims & FR, 2791885, A1 & JP, 2000-3	336021, A	1 2		
Y	WO, 97/19998, Al (NOVONORDISK AS 05 June, 1997 (05.06.97), Full text & AU, 7691296, A & EP, 86395 & JP, 2000-503042, A	3),	1-2		
¥	JP, 8-175935, A (Riaru Kagaku K 09 July, 1996 (09.07.96), Claim 1 (Family: none)	.K.),	1-2		
	and dominants are listed in the continuation of Box C.	See patent family annex.			
**Special categories of cited documents: "A" document defining the general state of the art which is not considered to be of particular relevance considered to be of particular relevance atter document but published on or after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention cannot considered to exhibit the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later "C" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention cannot considered novel or cannot be considered novel or ca		the application on the text of deletiving the invention cannot be dered to involve an inventive included invention cannot be the claimed invention cannot be the when the document is such documents, such soon skilled in the art on family			
Date of th	Date of the actual completion of the international search 22 February, 2001 (22.02.01) Date of mailing of the international search report 06 March, 2001 (06.03.01)		earch report .03.01)		
Name and	I mailing address of the ISA/ panese Patent Office	Authorized officer			
Facsimile No. Telephone No.					

Form PCT/ISA/210 (second sheet) (July 1992)

PATENT SPECIFICATION





Date of Application and filing Complete Specification: April 16, 1957. No. 12361/57.

Application made in Germany on Aug. 14, 1956.

Complete Specification Published: Feb. 4, 1959.

Index at acceptance:—Classes 2(2), B2B2; 2(4), P1D3, P9A(3A2:4F); 2(6), P8A, P8C(10:14B), P8D(2A:2B2:3A:4), P8K(4:8:10:11), P8T2C; and 15(2), B2L(1:5D).

International Classification: -- C08f. C09b. D01f. D06p.

COMPLETE SPECIFICATION

Improvements in the Dyeing of Fibres and like Structures of Polyacrylonitrile and Copolymers of Acrylonitrile

We, Badische Anilin- & Soda-Fabrik Aktiengesellschaft, a Joint Stock Company organised under the laws of Germany, of Ludwigshafen on Rhine, Germany, do hereby declare the invention, for which we pray that a patent may be granted to us, and the method by which it is to be performed, to be particularly described in and by the following statement:—

O This invention relates to an improved process for dyeing fibres and like structures of polyacrylonitrile and copolymers of acrylonitrile.

We have found that fibres and like structures
'15 of acrylonitrile polymers, i.e. polyacrylonitrile
and copolymers of acrylonitrile, can be dyed
in very fast, clear shades by using as dyestuffs watersoluble basic azo dyestuffs which
are resonance-hybrids between the general
to formulae Ia and Ib

or the corresponding colour bases, wherein m and n are zero or 1, X and Y are divalent atoms or groups which make up the nitrogenous rings to five- or six-membered rings, R and R^1 are alkyl, aralkyl, aryl or cycloalkyl and R^1 may also be hydrogen, and Z^θ represents the equivalent of an inorganic, organic or complex anion. Five- or six-membered rings of the kind indicated by the above general formulae are, for instance, thiazole, imidazole, pyrrolenine and pyridine rings and their benzo- and naphtho-derivatives.

Dyestuffs of the above general formulae

35 Ia is Ib can be prepared by various known methods for which no claim is made in the present application. Thus for example according to Fuchs and Graucus (ber. deutsch. Chem. [Price 3s. 6d.]

Ges. Band 61 (1928), page 63) heterocyclic hydrazones of the general formula II can be condensed with heterocyclic aldehydes of the general formula III or their anils, salts or quaternisation products:

the resultant condensation products being aftertreated if desired with quaternising agents, such as methyl iodide, dimethyl sulphate, toluene sulphonic acid ethyl ester, benzyl chloride, cyclohexyl bromide or cyclopentyl iodide.

Dyestuffs of the general formulae Ia > Ib are also obtained in many cases by diazotisation of heterocyclic amines of the formula IV and coupling with methylene bases of the formula V;

the condensation products again being aftertreated with quaternising agents if desired.

A further method for obtaining dyestuffs of the general formulae Ia > Ib consists in condensing hydrazines corresponding to the amines IV with aldehydes of the formula III and if desired quaternising the condensation products.

The basic azo dyestuffs of the formulae Ia ib accessible in these various ways are used according to this invention for dyeing flocks, fibres, threads, bands, woven or knitted fabrics of pure polyacrylonitrile or of copolymers of acrylonitrile with up to about 50% of other vinyl compounds, such as vinyl chloride, vinylidene chloride, vinyl fluoride,

Price 4s 6d

Price 33p

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vinyl acetate, vinyl propionate, vinylpyridine, vinylimidazole, vinyl alcohol, acrylic acid esters, methacrylic acid esters and/or acrylamide or methacrylamide.

The dyebaths are exhausted best in the acid pH range. It is possible however to dye in neutral to slightly alkaline baths; in this case it is often advantageous to lower the pH value during the dyeing process by adding small amounts of acid.

If instead of the salt-like dyestuffs of the formulae Ia lb, there are used the corresponding, often almost colourless, colour bases or anhydro bases, it has surprisingly been found that they go onto the fibres in the shade of colour of the intensely coloured dyestuff salts Ia lb. The dyeings do not change their shade of colour when given an alkaline washing.

The most favourable dyeing temperatures are somewhat different depending on the goods being dyed. In general the goods to be dyed are entered at about 40 to 60°C. and dyeing effected at the boiling temperature; it is also possible however to dye under static pressure at more than 100°C. up to about 135°C. The co-employment of the usual dyeing assistants is sometimes advantageous, but usually is unnecessary.

The dyestuffs can also be added to spinning solutions which contain polyacrylonitrile or copolymers of acrylonitrile with the above named other vinyl compounds. The spun-dyed structures thus obtained can if desired be further dyed or shaded in an aqueous bath with any dyestuffs having affinity to polyacrylonitrile, preferably the dyestuffs herein specified.

The dyeings obtained are characterised by lively, powerful shades and very good fastness, in many cases especially by excellent fastness to light.

The following Examples will further illustrate this invention but the invention is not restricted to these Examples. The parts specified in the Examples are parts by weight.

EXAMPLE 1.

0.5 part of the basic azo dyestuff of the formulae

is dissolved in 4,000 parts of water with the addition of 5 parts of 30% acetic acid. 100 parts of a fabric of polyacrylonitrile staple fibre is entered at 50°C, heated to boiling in 30 minutes and kept at the boiling temperature for an hour. The fabric is then rinsed and dried. A powerful, reddish-yellow dyeing of very good fastness to moisture and excellent fastness to light is obtained.

Fast dyeings are obtained in the same way with the following dyestuffs (of which but one limiting formula of the resonating system is indicated):—

•

EXAMPLE 2.
100 parts of fibrous material of a copolymer prepared from 95% of acrylonitrile and

5% of butyl-methacrylate are dyed in a bath containing in 5,000 parts of water 0.8 parts of the basic azo dyestuff of the formulae:

and 2 parts of chloracetic acid. The fibrous
 material is dyed a very fast and powerful redorange.

The following dyestuffs (of which but one limiting formula is indicated) can be used in an analogous way:

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$$\begin{bmatrix} CH_{3}C + N = N - CH = C + CH_{3} \\ CH_{3} \end{bmatrix} = CH^{S} - N = N - CH = C + CH_{3} \\ CH_{3} \end{bmatrix} = CH^{S} - N = N - CH = C + CH_{3} \\ CH_{3} \end{bmatrix} = CH_{3}SO_{4}$$

$$\begin{bmatrix} CH_{3}C + N = N - CH = C + CH_{3} \\ CH_{3} \end{bmatrix} = CH_{3}SO_{4}$$

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$$\begin{bmatrix} CH_{3}C + N = N - CH = C + CH_{3} \\ CH_{3} \end{bmatrix} = CH_{3}SO_{4}$$

$$\begin{bmatrix} CH_{3}C + N = N - CH = C + CH_{3} \\ CH_{3} \end{bmatrix} = CH_{3}SO_{4}$$

$$\begin{bmatrix} CH_{3}C + N = N - CH = C + CH_{3} \\ CH_{3} \end{bmatrix} = CH_{3}SO_{4}$$

$$\begin{bmatrix} CH_{3}C + N = N - CH = C + CH_{3} \\ CH_{3} \end{bmatrix} = CH_{3}SO_{4}$$

$$\begin{bmatrix} CH_{3}C + N = N - CH = C + CH_{3} \\ CH_{3} \end{bmatrix} = CH_{3}SO_{4}$$

$$\begin{bmatrix} CH_{3}C + N = N - CH = C + CH_{3} \\ CH_{3} \end{bmatrix} = CH_{3}SO_{4}$$

$$\begin{bmatrix} CH_{3}C + N = N - CH = C + CH_{3} \\ CH_{3} \end{bmatrix} = CH_{3}SO_{4}$$

$$\begin{bmatrix} CH_{3}C + N = CH_{3} \\ CH_{3} \end{bmatrix} = CH_{3}SO_{4}$$

$$\begin{bmatrix} CH_{3}C + N = CH_{3} \\ CH_{3} \end{bmatrix} = CH_{3}SO_{4}$$

$$CH_{3}C + CH_{3}C + CH_{3}C$$

$$CH_{3}C + CH$$

Example 3. Polyacrylonitrile fibres are dyed as in Example 1 with a bath containing, in 4,000. parts of water, 5 parts of 30% acetic acid and 5 0.5 part of the dyestuff of the formulae:

or the corresponding anhydro base of the formula:

10 The fibres are dyed a powerful greenish, yellow; the dyeings have very good fastness to light.

In a similar way there may be used for

dyeing the following anhydro bases or their hydrochlorides, sulphates, oxalates or chloracetates:

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WHAT WE CLAIM IS:-

1. A process for dyeing fibres or like structures of polyacrylonitrile or copolymers of acrylonitrile with basic dyestuffs wherein there is used as dyestuff a basic azo dyestuff which is a resonance hybrid between the general formulae Ia and Ib

or the corresponding colour base, in which m and n are zero or 1, X and Y represent divalent atoms or groups which make up the nitrogenous rings to five or six-membered rings, R and R¹ represent alkyl, aralkyl, aryl
or cycloalkyl and R¹ may also be hydrogen, and Z^θ is the equivalent of an inorganic, organic or complex anion.

2. The process for the dyeing of fibres of polyacrylonitrile or copolymers of acrylonitrile substantially as described in any of the foregoing Examples.

3. Fibres and like structures of polyacrylonitrile or copolymers of acrylonitrile when dyed by the process claimed in claim 1 or 2.

4. Fibres or like structures of polyacrylonitrile or copolymers of acrylonitrile coloured with basic dyestuffs which are resonance hybrids between the general formulae Ia and Ib

or the corresponding colour bases, in which m and n are zero or 1, X and Y represent divalent atoms or groups which make up the nitrogenous rings to five- or six-membered rings, R and R^1 represent alkyl, aralkyl, aryl or cycloalkyl and R^1 may also be hydrogen, and Z^θ is the equivalent of an inorganic, organic or complex anion.

J. Y. & G. W. JOHNSON,47, Lincoln's Inn Fields, London, W.C.2,Chartered Patent Agents.

Learnington Spa: Printed for Her Majesty's Stationery Office, by the Courier Press.—1959. Published by The Patent Office, 25, Southampton Buildings, London, W.C.2, from which copies may be obtained.

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